

SINCE 1984  **Brilliant**[®]
STUDY CENTRE, PALA

JEE MAIN 2026

SESSION-2

SHIFT-1 MORNING



SCAN ME

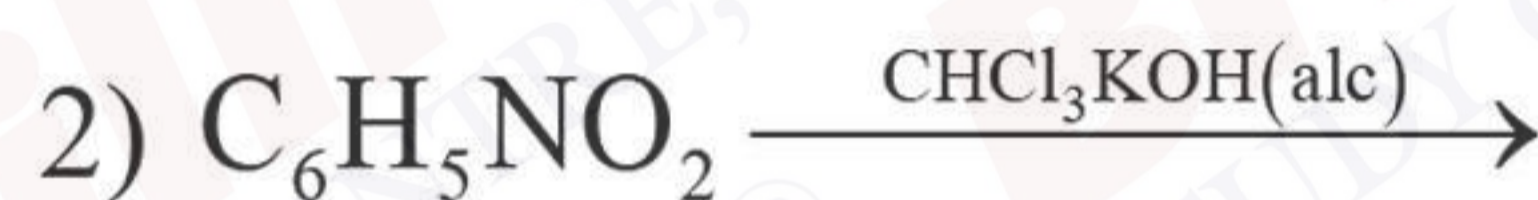
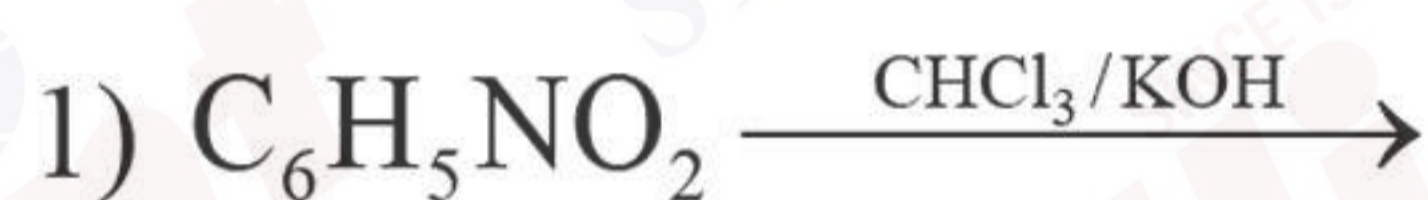
VIDEO SOLUTION

MEMORY BASED QUESTIONS

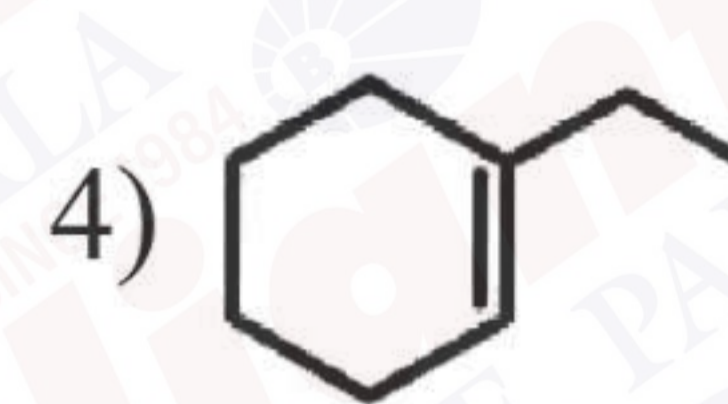
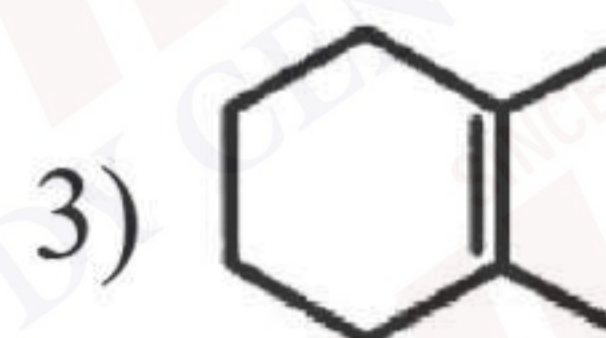
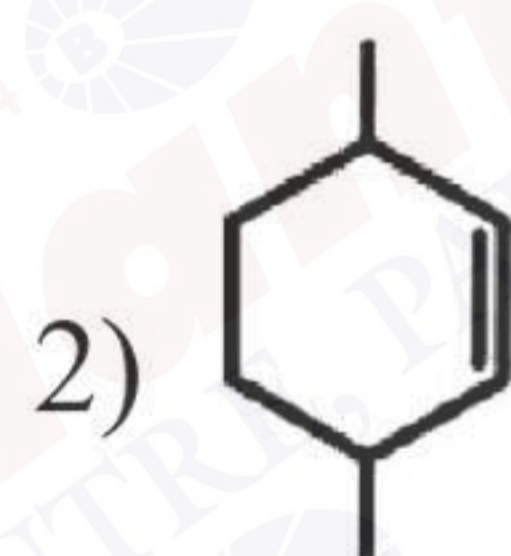
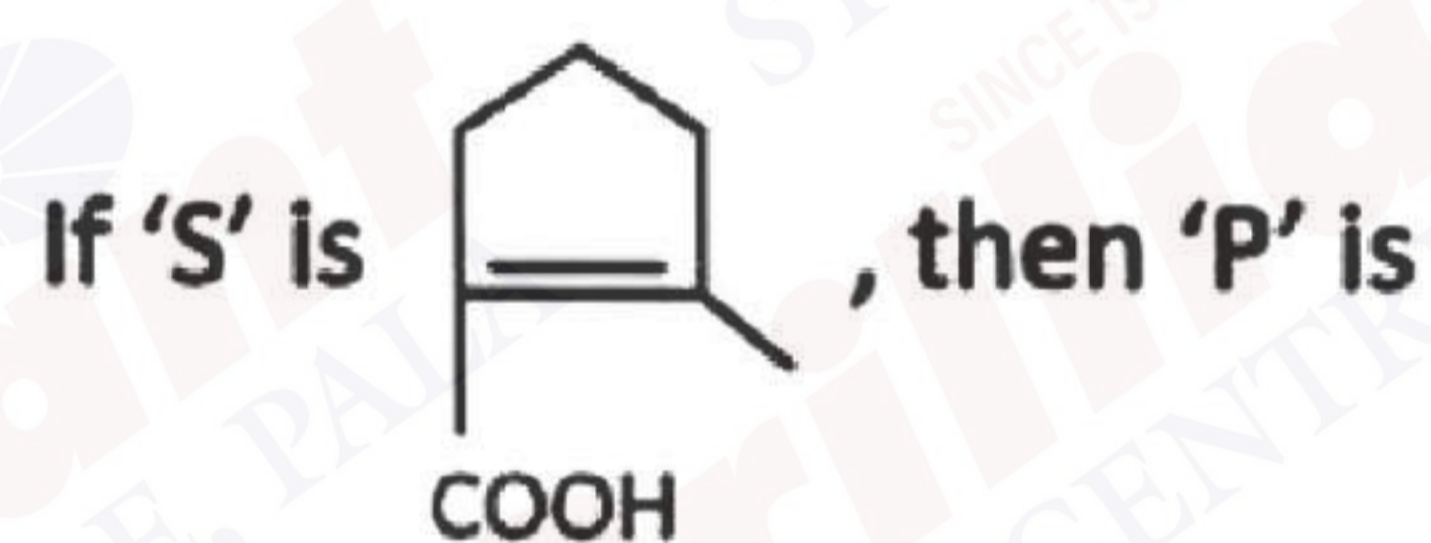
- 0.2 g of organic compound is subjected to estimation of 'S' by Carius method, giving 0.6 g BaSO₄. Find % of S. (Nearest integer)
- Compare the energy of orbitals for multielectronic species.

	n	l	m
A	3	0	0
B	3	1	-1
C	4	2	0
D	3	2	1

- 1) C > D > B > A 2) C > B > D > A 3) A > B > C > D 4) A > B > D > C
- Half life for a first order reaction is 6.93 min. What is the time required (in min) to complete 90% of reaction? (Nearest integer)
 - Which of the following will produce C₆H₅NC?



- How many compounds among the following having sp³d hybridisation of central atom BrF₅, XeF₅⁻, ICl₂⁻, ICl₄⁻, SF₄, NH₄⁺, ClF₃, XeF₂, XeF₄
- Consider the sequence:



- Given below are two statements:

Statement I : K₂Cr₂O₇ can be used as a primary standard

Statement II : Phenolphthalein is a weak base indicator

In the light of above statements choose the correct option.

- Both statement I and statement II are correct
- Statement I is correct but statement II is incorrect
- Statement I is incorrect but statement II is correct
- Both statement I and statement II are incorrect

8. Consider the following reaction:

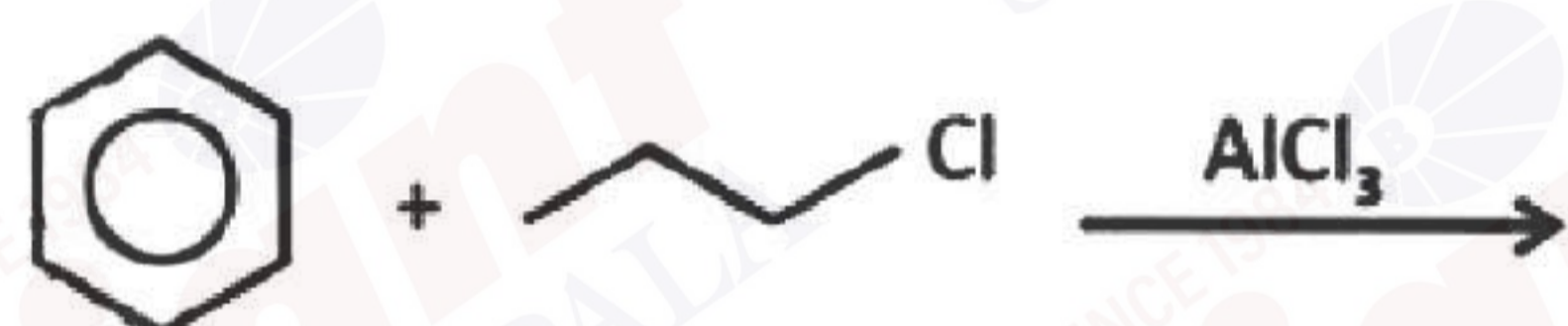


Change in % of oxygen from phenol and product in filtrate is x%. Value of 2x is

9. What amount of residual will be produced on heating 2.76 gram of Ag_2CO_3 ?

- 1) 1.08 g 2) 2.16 g 3) 3.2 g 4) 4.32 g

10. Consider the following reaction



Choose the incorrect statement

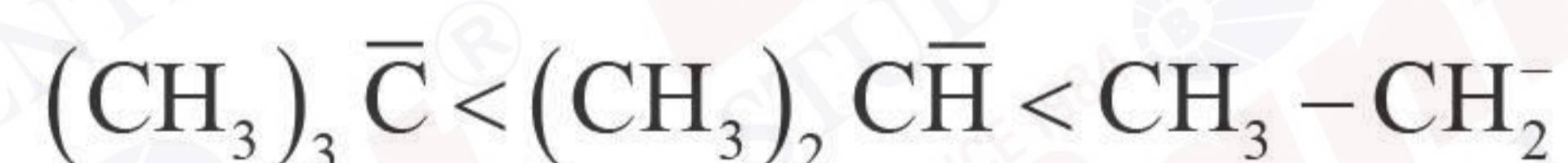
- 1) Isopropylation intermediate is formed in the reaction
- 2) Isopropyl benzene is the major product formed
- 3) Rearrangement of carbocation occurs
- 4) The product is less reactive than benzene towards electrophilic aromatic substitution

11. pH of 10^{-7} M aq. KOH solution at 25° is,

- 1) 6.50 2) 6.28 3) 7.02 4) 7.00

12. Given below are two statements:

Statement I : stability of carbanion is as follows



Statement II : Stability of carbanion can be explained on the basis of inductive effect

- 1) Both statement I and II are correct
- 2) Both statement I and II are incorrect
- 3) Statement I is correct and statement II is incorrect
- 4) Statement I is incorrect and statement II is correct

13. Given below two statements:

Statement I : In octahedral complexes each electrons in t_{2g} orbital stabilises by $-0.4 \Delta_0$ and that of e_g orbital destabilises by $+0.6\Delta_0$

Statement II : All d-electrons are of same energy but after complex is formed its degeneracy is disturbed according to crystal field theory.

- 1) Both statement I and statement II are correct
- 2) Both statement I and statement II are incorrect
- 3) Statement I is correct, statement II is incorrect
- 4) Statement I is incorrect, statement II is correct

14. Consider the following statements

- a) Glucose exists in two anomeric form
- b) Melting point of α -anomer is greater than β -anomer
- c) Specific rotation of α -anomer is 19° and for β -anomer is 112°
- d) α and β anomers are made at temperature 303 K and 371 K

- 1) a,b,c only 2) a,d only 3) a,c,d only 4) a,b,c,d

15. Statement I : Order of electronegativity is $F > O > N$

Statement II In OF_2 , oxidation state of 'O' is +2 and in N_2O , the value of oxidation state of O is -2

- 1) Both statement I and statement II are wrong
- 2) Statement II is wrong, statement I is correct
- 3) Both statement I and statement II are correct
- 4) Statement I is correct, statement II is wrong

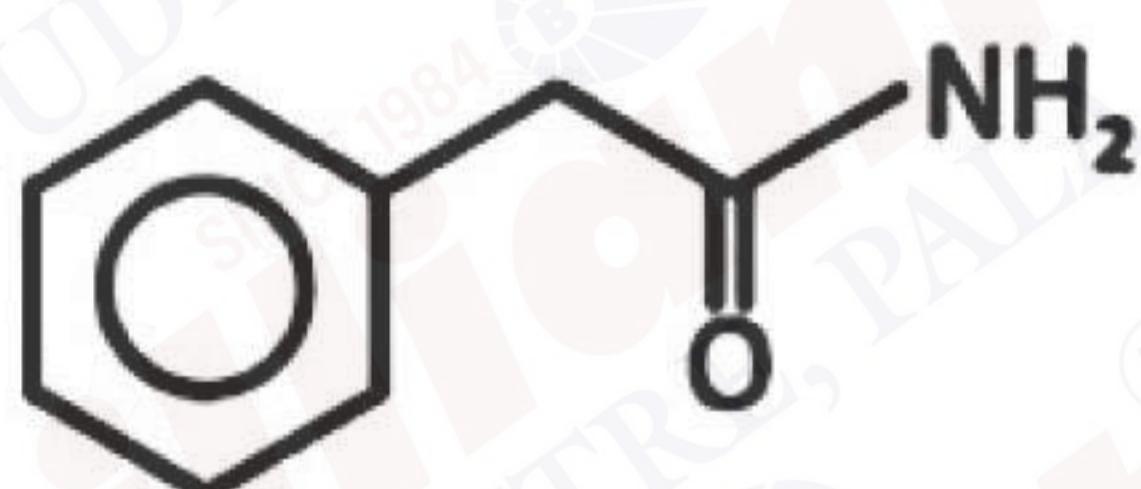
16. Arrange the following mixture in increasing order of pH at $25^\circ C$ (each solution is decimolar).

- a) 10 mL of HCl + 10 mL of $Ca(OH)_2$
- b) 10 mL of HCl + 25 mL of $Ca(OH)_2$
- c) 10 mL of HCl + 10 mL of H_2SO_4

- 1) $b < a < c$ 2) $a < b < c$ 3) $c < a < b$ 4) $b < c < a$

17. More fraction of H_2O in 10% $\frac{W}{W}$ solution of urea in water is $x \times 10^{-3}$. Find value of x.

18. Consider the following statements:

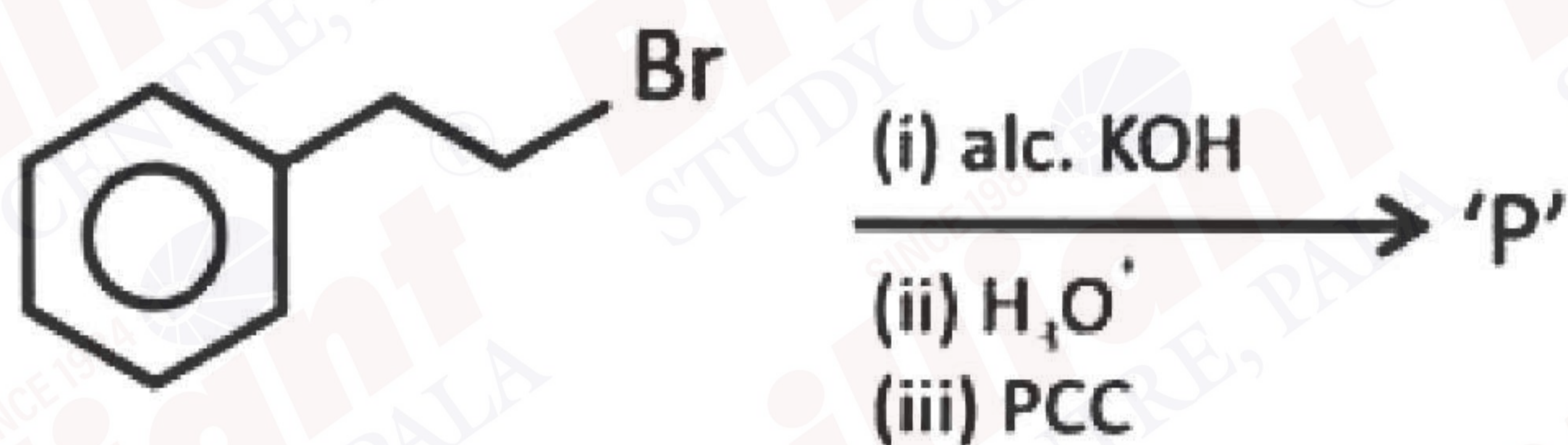
Statement I :  \xrightarrow{NaOBr} primary aromatic amine

Statement II :  can be prepared by Gabriel phthalimide synthesis

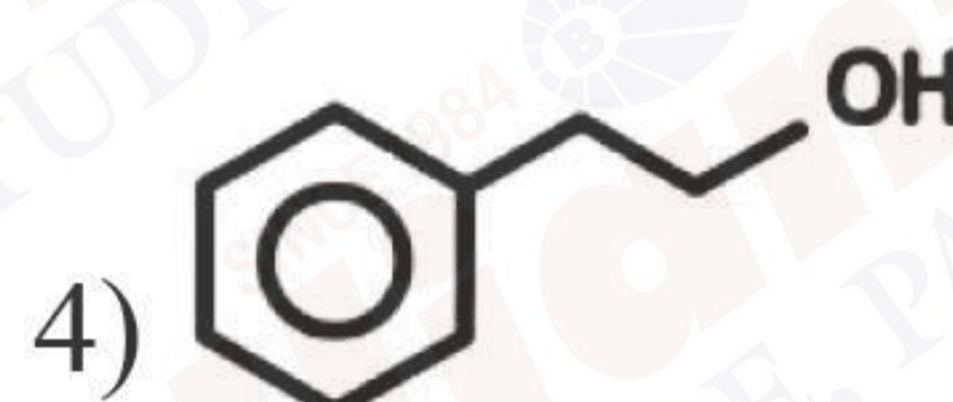
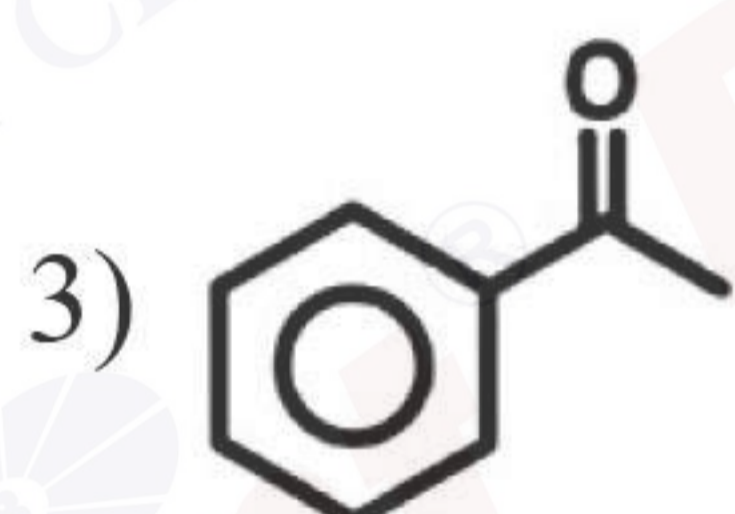
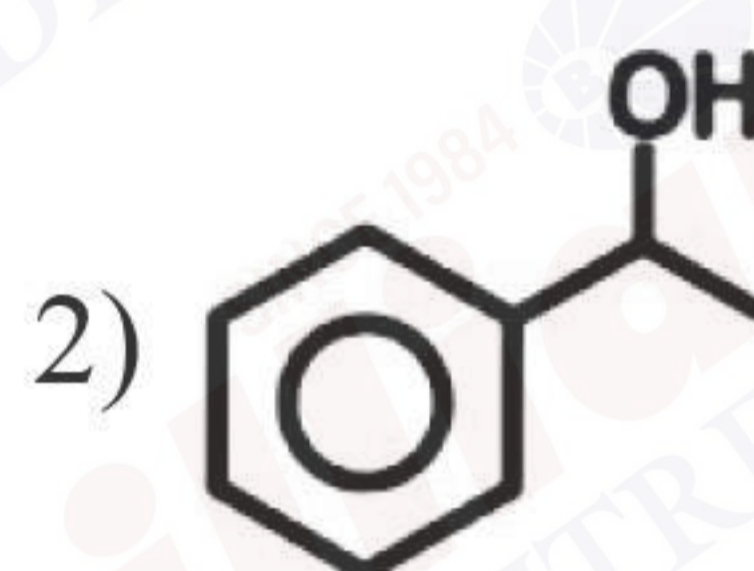
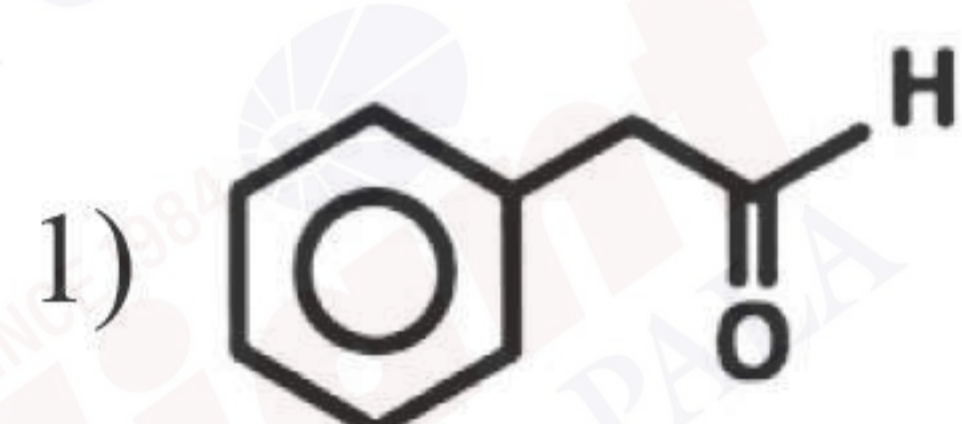
Choose the correct option

- 1) Both statement I and statement II are correct
- 2) Statement I is correct but statement II is incorrect
- 3) Statement I is incorrect, statement II is correct
- 4) Statement I and statement II are incorrect

19. Consider the following sequence of reactions



The major final product (P) is



20. The correct relationship between the molar concentration of the anion $[A^{3-}]$ and the solubility product constant (K_{sp}) for an M_3A_2 type salt

1) $[A^{3-}] = \left(\frac{K_{sp}}{108}\right)^{\frac{1}{5}}$
 2) $[A^{3-}] = \left(\frac{8 K_{sp}}{27}\right)^{\frac{1}{5}}$
 3) $[A^{3-}] = \left(\frac{27 K_{sp}}{8}\right)^{\frac{1}{5}}$
 4) $[A^{3-}] = \left(\frac{4 K_{sp}}{9}\right)^{\frac{1}{5}}$

21. The order of acidic strength for 0.1 M aqueous solution of following is



1. The sum of value $\sum_n^{10} = 1 \frac{528}{n(n+1)(n+2)}$ is equal to

2. The value of $\int_0^\infty \frac{\log x}{x^2 + 4} dx$ is equal to

3. Consider an Equilateral ΔPQR . where $P(3,5)$ & \overline{QR} is $x+y=4$. If the orthocenter of ΔPQR is (α, β) . Then $9(\alpha + \beta) =$

4. The area enclosed between the region given by $xy \leq 27$ & $1 \leq y \leq x^2$ is

- 1) $54 \ln 3 - \frac{12}{3}$ 2) $52 \ln 3 - \frac{52}{2}$ 3) $54 \ln 2 - \frac{14}{2}$ 4) $52 \ln 2 - \frac{52}{1}$

5. Consider a set $A = \{1, 2, 3, 4, 5, 6\}$. The number of one-one functions $f : A \rightarrow A$ such that $f(1) \geq 3, f(3) \leq 4, f(2) + f(3) = 5$ is equal to

- 1) 144 2) 72 3) 36 4) 24

6. Let S_n is the sum of first n terms of an A.P. If $S_n = 3n^2 + S_n$. Then the sum of square of first ten terms of the given P is

- 1) 15110 2) 15220 3) 14202 4) 14303

7. The square of distance of point $P(5,6,7)$ from the line $\frac{x-2}{2} = \frac{y-5}{3} = \frac{z-2}{4}$ is equal to

8. If $\tan A$ and $\tan B$ are the roots of the equation $x^2 - 2x - 5 = 0$. Then, the value of $20 \sin^2 \left(\frac{A+B}{2} \right)$ is

- 1) $5 + 3\sqrt{6}$ 2) $10 + 3\sqrt{10}$ 3) $5 - 3\sqrt{2}$ 4) $10 - 3\sqrt{10}$

9. The value of $\int_{\frac{\pi}{6}}^{\frac{\pi}{3}} \frac{(4 - \operatorname{cosec}^2 x)}{\cos^4 x} dx$ is equal to

- 1) $\frac{16\sqrt{3}}{9}$ 2) $\frac{32\sqrt{3}}{9}$ 3) $\frac{16}{\sqrt{3}}$ 4) 32

10. Let $\vec{a} = \sqrt{7}\hat{i} + \hat{j} - \hat{k}$, $\vec{b} = \hat{j} + 2\hat{k}$, $\vec{r} \times \vec{a} + \vec{b} \times \vec{a} = 0$ and $\vec{r} \cdot \vec{a} = 0$. The value of $|\vec{r}|^2$ is equal to
 1) 56 2) 44 3) 42 4) 48
11. Consider an Ellipse E: $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ with eccentricity $e = \frac{4}{5}$, focus at (4,0) and the P(3, α) lie on E. Then the area of the triangle POS(in sq. unit) is
 1) $\frac{12}{5}$ 2) $\frac{13}{5}$ 3) $\frac{24}{5}$ 4) $\frac{48}{5}$
12. If $\alpha + \frac{\pi}{4} + \sum_{p=1}^{11} \tan^{-1} \left(\frac{2^{p-1}}{1+2^{2p-1}} \right)$, then the value of $\tan \alpha$ is
 1) 2^9 2) 2^{10} 3) 2^{11} 4) 2^{12}
13. Letter is posted from either KANPUR or ANANTPUR. When envelope was open only "AN" was visible. The probability that it came from ANANTPUR is equal to
 1) $\frac{9}{16}$ 2) $\frac{1}{11}$ 3) $\frac{10}{17}$ 4) $\frac{9}{13}$
14. The sum of integral values of P for which the equation $3 \sin^2 x - 12 \cos x - 3 = P$ has at least one zero, is
 1) -60 2) -75 3) -90 4) 110
15. Let p(x,y) is a variable point on the circle $x^2 + y^2 - 6x - 8y + 21 = 0$, then the maximum possible distance of p from the vertex of $y^2 + 6y + 13 = 0$ is
 1) $7 + 2\sqrt{2}$ 2) $2 + 7\sqrt{2}$ 3) $4 + 7\sqrt{2}$ 4) $3 + 2\sqrt{2}$
16. If α, β are the roots of equation $ax^2 + bx + c = 0$ $|\alpha - \beta| = \sqrt{11}$, $\alpha + \beta = 3i$, then find the value of $(\alpha^3 + \beta^3)^2$ is
17. If $\lim_{x \rightarrow 0} \frac{1 - \cos ax \cos((a+1)x) \cos((a+2)x)}{\sin((a+1)x)^2} = 2$. Then the product of all possible values of a is
 1) 1 2) -1 3) 2 4) -2
18. Let $S_n = \left(\frac{1}{x^3} - x^4 \right)^n$, the sum of coefficients of x^7 & x^{14} is zero. Then find the value of n is
19. Two players A and B play a series of matches. The winner will be the one who wins 5 matches. Find the number of ways in which A can win-----

20. If $x - 2y + tz = 0$

$$3x + 5y + t^2z = 0$$

$$6x + ty + f(t)z = 0$$

$$D = 0$$

$$\begin{vmatrix} 1 & -2 & t \\ 3 & 5 & t^2 \\ 6 & + & f(t) \end{vmatrix} = 0$$

has infinitely many solutions then

1) $y = f(t)$ is strictly increasing

2) $y = f(t)$ is strictly decreasing

3) $y - f(t)$ is decreasing

4) $y - f(t)$ is increasing

Consider

$$\begin{matrix} x_i & 5 & 6 & 8 & 11 & 13 \\ f_i & 4 & 8 & 2 & 3 & 9 \end{matrix} \rightarrow 26$$

then mean deviation about mean is

1) 4.23

2) 5.23

3) 2.32

4) 3.23

10. In a region where electric field exist as $-E_0 \cdot (\text{V/m})$. Initial (at $t=0$) velocity of particle of mass

m is $4V_0$, $\lambda_0 = \frac{h}{4mv_0}$ at instant $t = 0$, then find λ in terms of λ_0 at time instant t .

- 1) $\lambda = \frac{h}{h + qE_0\lambda_0 t}$ 2) $\lambda = \frac{h\lambda_0}{h - qE_0\lambda_0 t}$ 3) $\lambda = \frac{h\lambda_0}{(h + 2E_0q\lambda_0 t)}$ 4) $\lambda = \frac{h\lambda_0}{\left(h + \frac{qE_0\lambda_0 t}{2}\right)}$

11. Moment of inertia of rod about an axis passing through point at distance $\frac{\ell}{4}$ from centre and perpendicular to rod is (The uniform rod is of mass m and length ℓ)

- 1) $\frac{8}{63} m\ell^2$ 2) $\frac{3}{7} m\ell^2$ 3) $\frac{7}{44} m\ell^2$ 4) $\frac{7}{48} m\ell^2$

12. For a wave, equation is given as $y = 3 \sin(\omega t - 0.018x + \pi/4)$. Find the minimum distance between consecutive crest.

- 1) $\frac{200}{3} \pi(\text{m})$ 2) $\frac{4000}{3} \pi(\text{m})$ 3) $\frac{1000}{9} \pi(\text{m})$ 4) $\frac{500}{9} \pi(\text{m})$

13. In thermodynamics, with usual symbols two expressions are given

$$\text{Expression 1} = \Delta U = \mu C_v \Delta T + \frac{\mu R \Delta T}{\gamma - 1}$$

$$\text{Expression 2} = \gamma f = f + 2$$

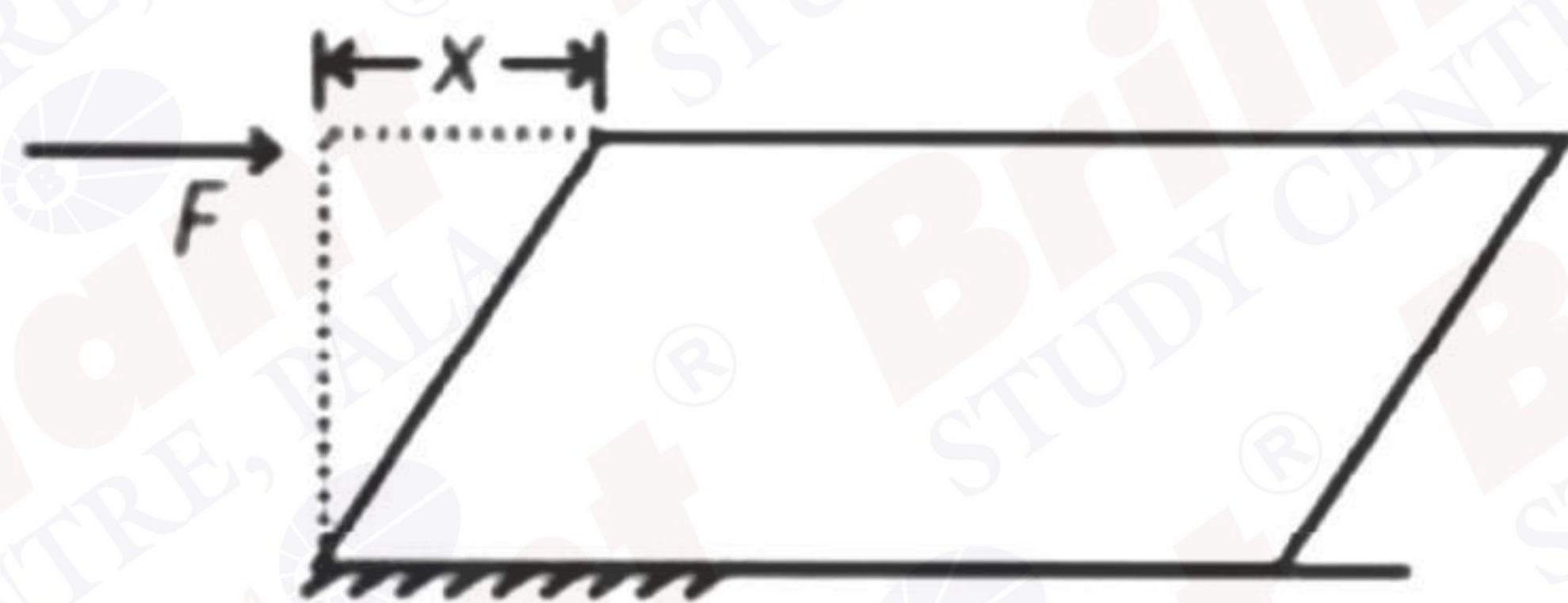
Then

- 1) E1 is correct and E2 is incorrect 2) E1 is incorrect and E2 is correct
 3) E1, E2 are correct 4) E1, E2 are incorrect

14. A ball is dropped from 18 m high tower. Find distance of ball from ground when the speed of ball becomes equal to magnitude of acceleration of ball ($g = 10 \text{ m/s}^2$)

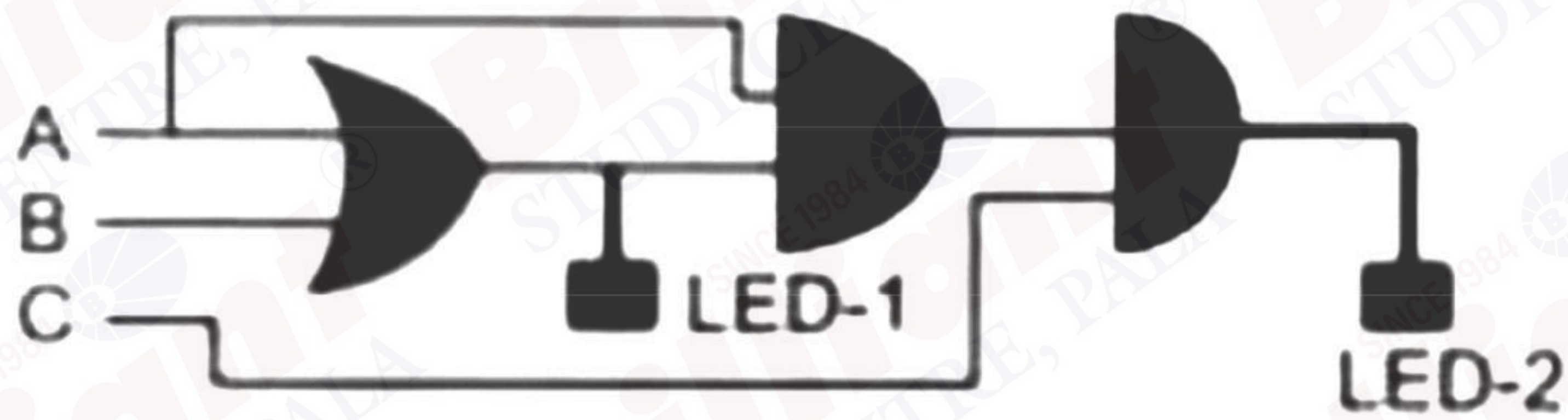
- 1) 8 2) 18 3) 13 4) 10

15. On the top surface of a cube (length = 5cm) force F is acting tangentially. Lower surface is fixed, then find shifting x . [Modulus of rigidity $\eta = 10^5$ (SI unit)]



- 1) $x = \frac{F}{2500}$ 2) $x = \frac{F}{5000}$ 3) $x = \frac{F}{4000}$ 4) $x = \frac{F}{8000}$

16. Both LED's will switch ON in which of the following option



1)

A	B	C
1	1	0

2)

A	B	C
1	0	0

3)

A	B	C
1	0	1

4)

A	B	C
0	1	1

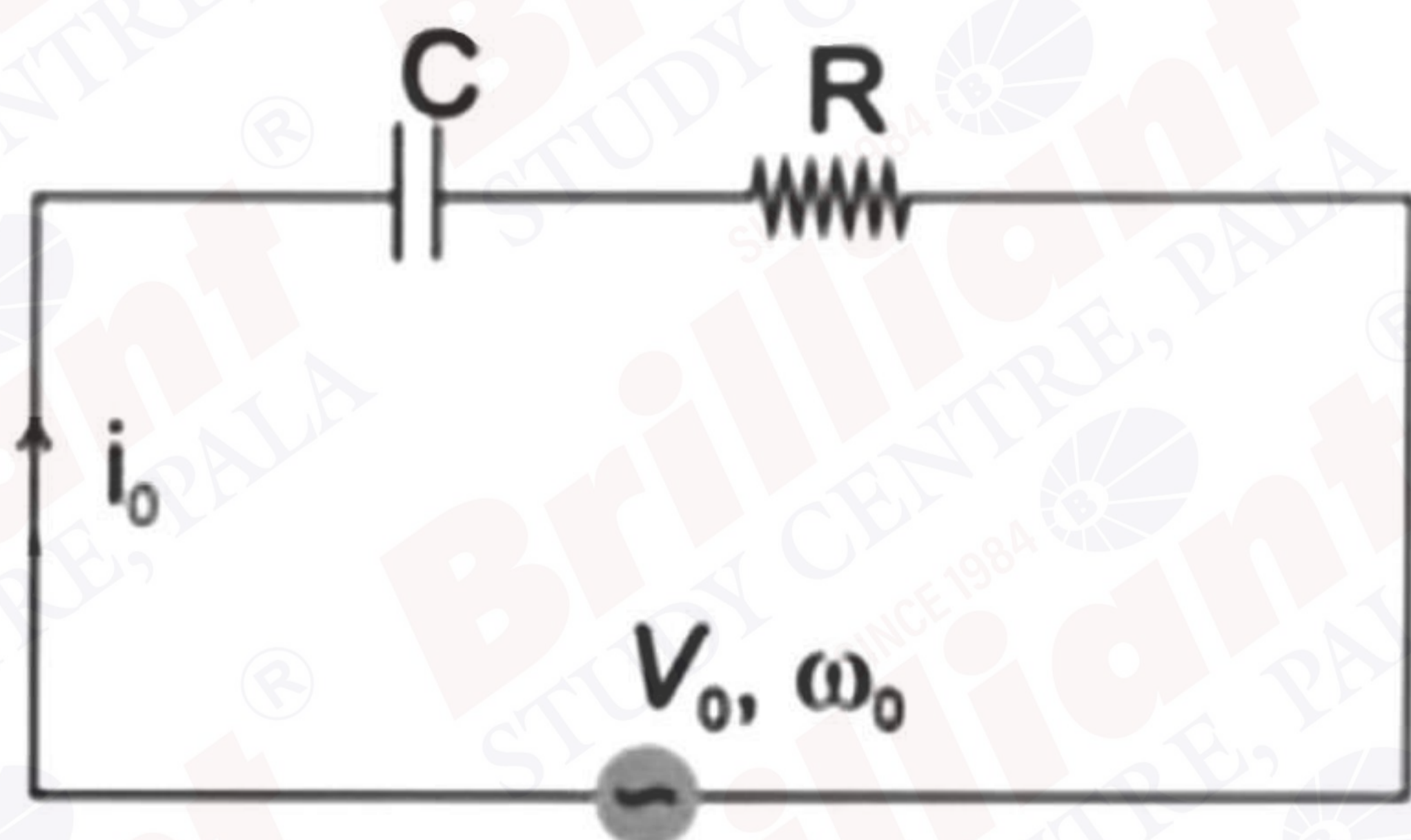
17. VSD matches with 9 MSD in vernier calliper. 1 MSD is of 1 mm. When no reading is taken, the zero of vernier scale is just right of zero of main scale and 7th VSD matches with one MSD then zero error is

- 1) 0.7 cm, +ve 2) 0.7 cm, -ve 3) 0.7 mm, -ve 4) 0.7 mm, +ve

18. Hydrogen atom an electron is revolving in 4th Bohr's orbit. Find wavelength of the incident radiation to excite this electron to 16th Bohr's orbit. [Use $hc = 1240 \text{ nm. eV}$]

- 1) 1556 nm 2) 400 nm 3) 800 nm 4) 2000 nm

19. In circuit given below, current in circuit is i_0 . If angular frequency is changed from ω_0 to $\omega_0/8$, the current becomes $i_0/4$. Find $\frac{1}{\omega_0 CR}$.



- 1) $\sqrt{\frac{5}{16}}$ 2) $\sqrt{\frac{42}{13}}$ 3) $\sqrt{\frac{51}{38}}$ 4) $\sqrt{\frac{3}{2}}$

20. A particle moving uniformly enters a region having uniform magnetic field $(3\hat{i} + 2\hat{j})\text{T}$. Its

acceleration is $\left(4\hat{i} + \frac{x}{2}\hat{j}\right)\text{m/s}^2$. Find the value of x.