

SINCE 1984  **Brilliant**[®]
STUDY CENTRE, PALA

JEE MAIN 2026

SESSION-1 SHIFT-1 MORNING

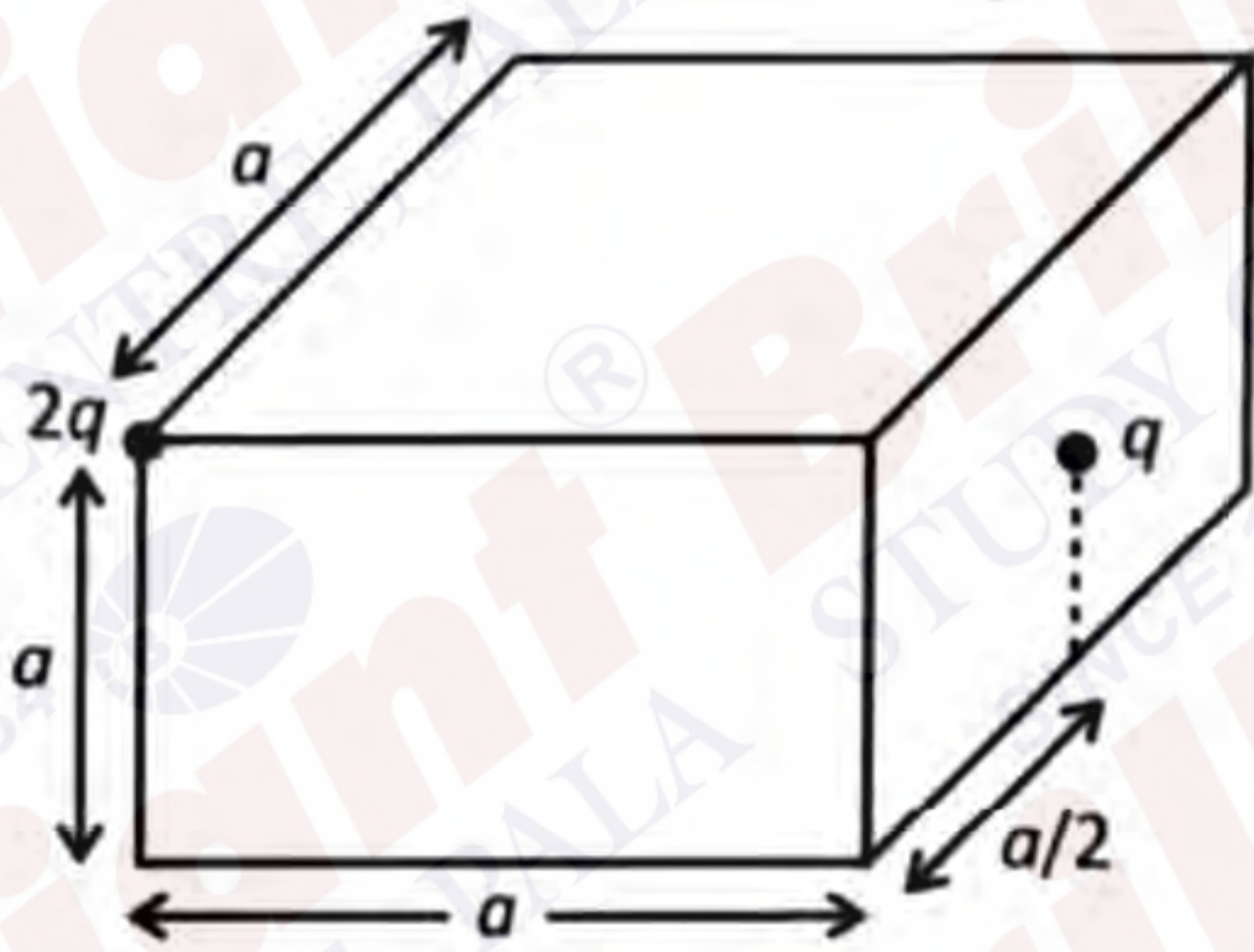


VIDEO SOLUTION
JANUARY 23, 2026
FRIDAY

SCAN ME

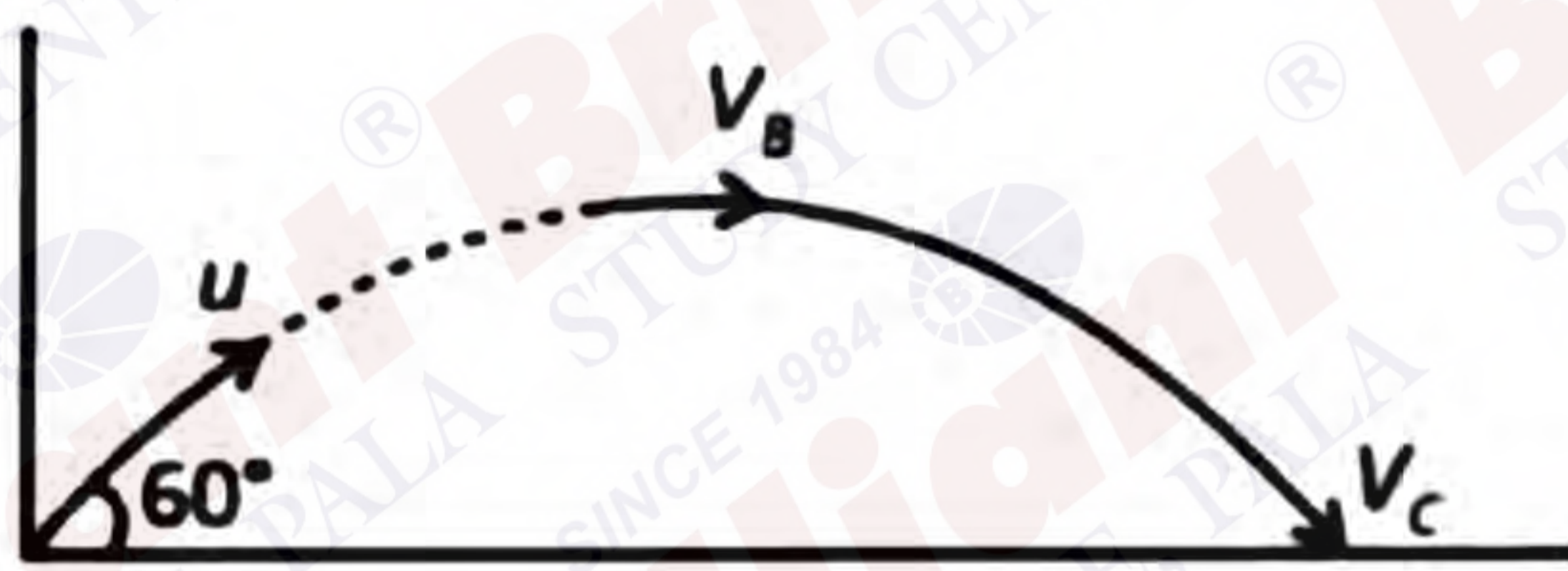
MEMORY BASED QUESTIONS

QN There are two point charges, one at vertex and other at face as shown the cube. Find electric flux through the cube.



- 1) $3q/\epsilon_0$ 2) q/ϵ_0 3) $3q/4\epsilon_0$ 4) $5q/\epsilon_0$

QN If a projectile is being projected with speed v at angle 60° with horizontal. Find the ratio of speed at highest point (v_B) to the speed at final point (v_C)



- 1) 3 : 4 2) 1 : 3 3) 1 : 2 4) 1 : 12

QN Find out the correct energy for the ground state or energy transition. (symbols has usual meaning & $n \rightarrow m$ gives the transition)

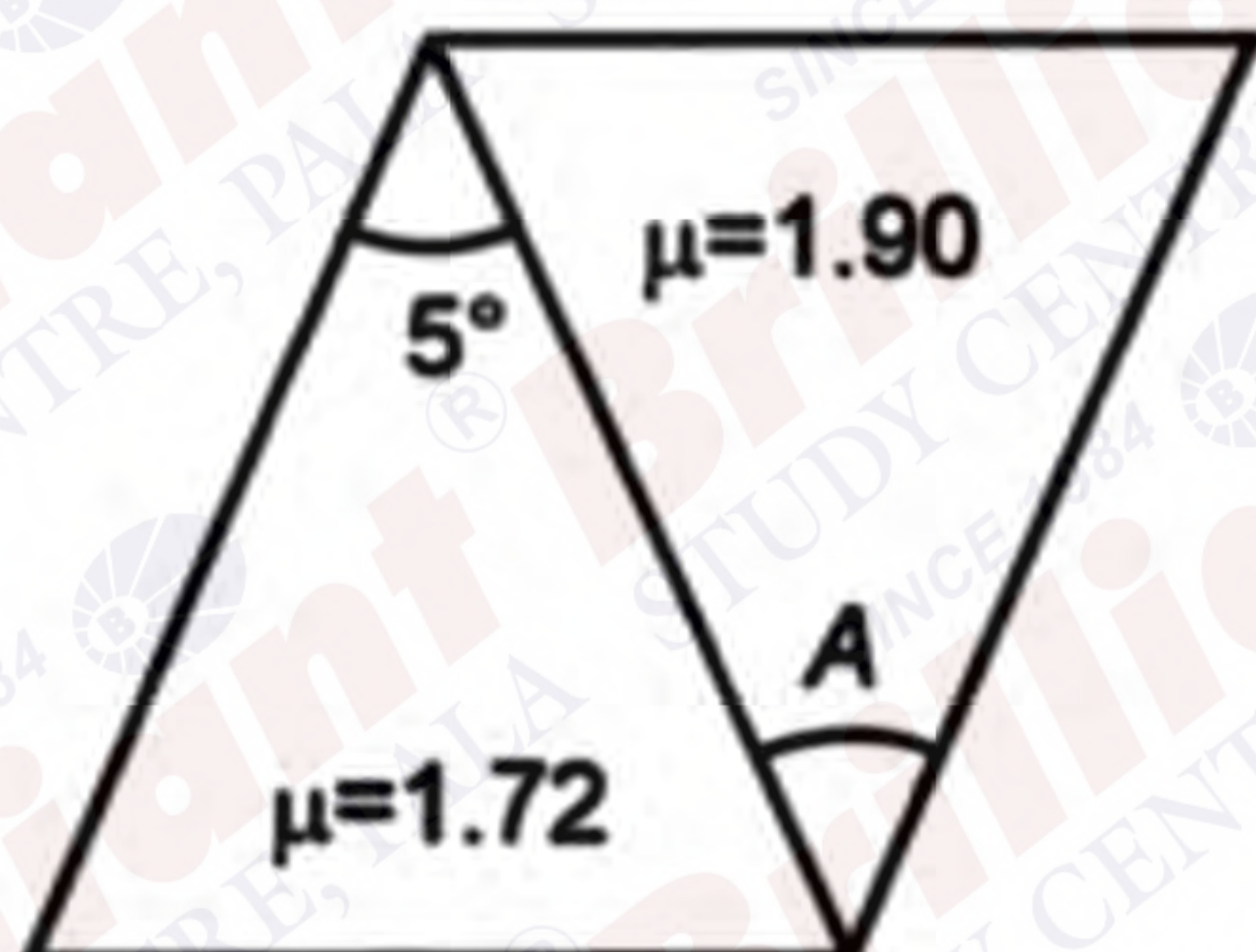
- 1) H (-6.8 eV) 2) Li^{2+} (-13.6 eV) 3) $He_{2 \rightarrow 1}^+$ (40.8 eV) 4) $Be_{2 \rightarrow 1}^{3+}$ (-13.6 eV)

QN For the given set of measurement find relative error.

20.00, 19.75, 18.25, 17.01

- 1) 0.12 2) 0.06 3) 0.09 4) 0.17

QN Find A for dispersion without deviation



- 1) 3 2) 4 3) 4.5 4) 5

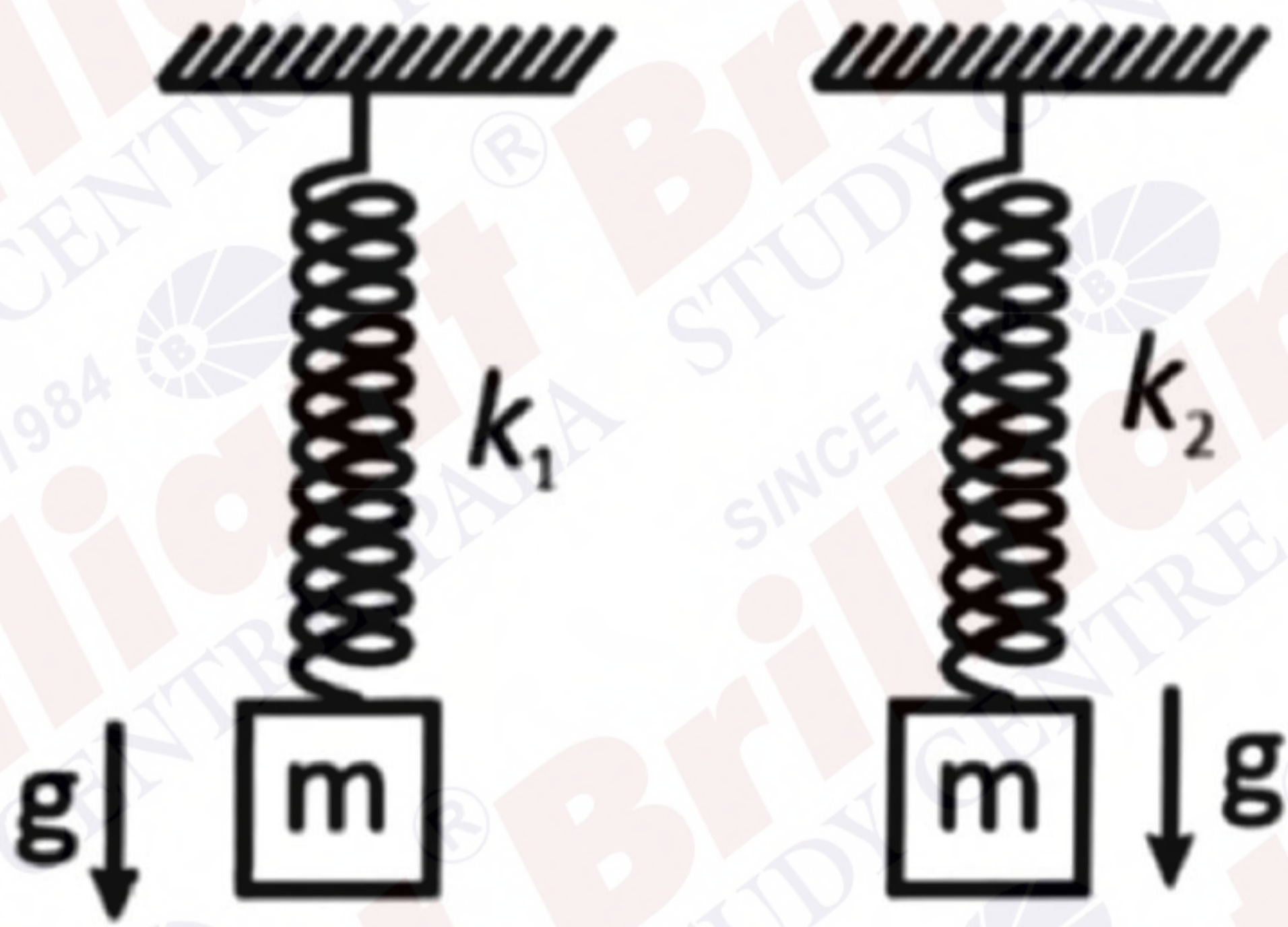
QN In two different YDSE setup for two different monochromatic wave are used but fringe width on the screen is same. If $\frac{\lambda_1}{\lambda_2} = \frac{1}{2}$ and $\frac{d_1}{d_2} = \frac{2}{1}$, then find the ratio of $\frac{D_1}{D_2}$ (all symbols are standard).

- 1) 1 2) 1/2 3) 2 4) 4

QN A 25 kg mass moving with 30 m/s and another mass 15 kg of same nature moving with 10 m/s in opposite direction, collides perfectly inelastically. Find the rise in temperature of the system. (Given $C = 5 \times \text{cal/g}^\circ\text{C}$ & $1 \text{ cal} = 4.2 \text{ J}$)

- 1) $1/50^\circ\text{C}$ 2) $1/112^\circ\text{C}$ 3) $1/150^\circ\text{C}$ 4) $1/125^\circ\text{C}$

QN Find ratio of energy stored in the two springs as shown in figure below ($2K_1 = K_2, 4M_1 = M_2$)

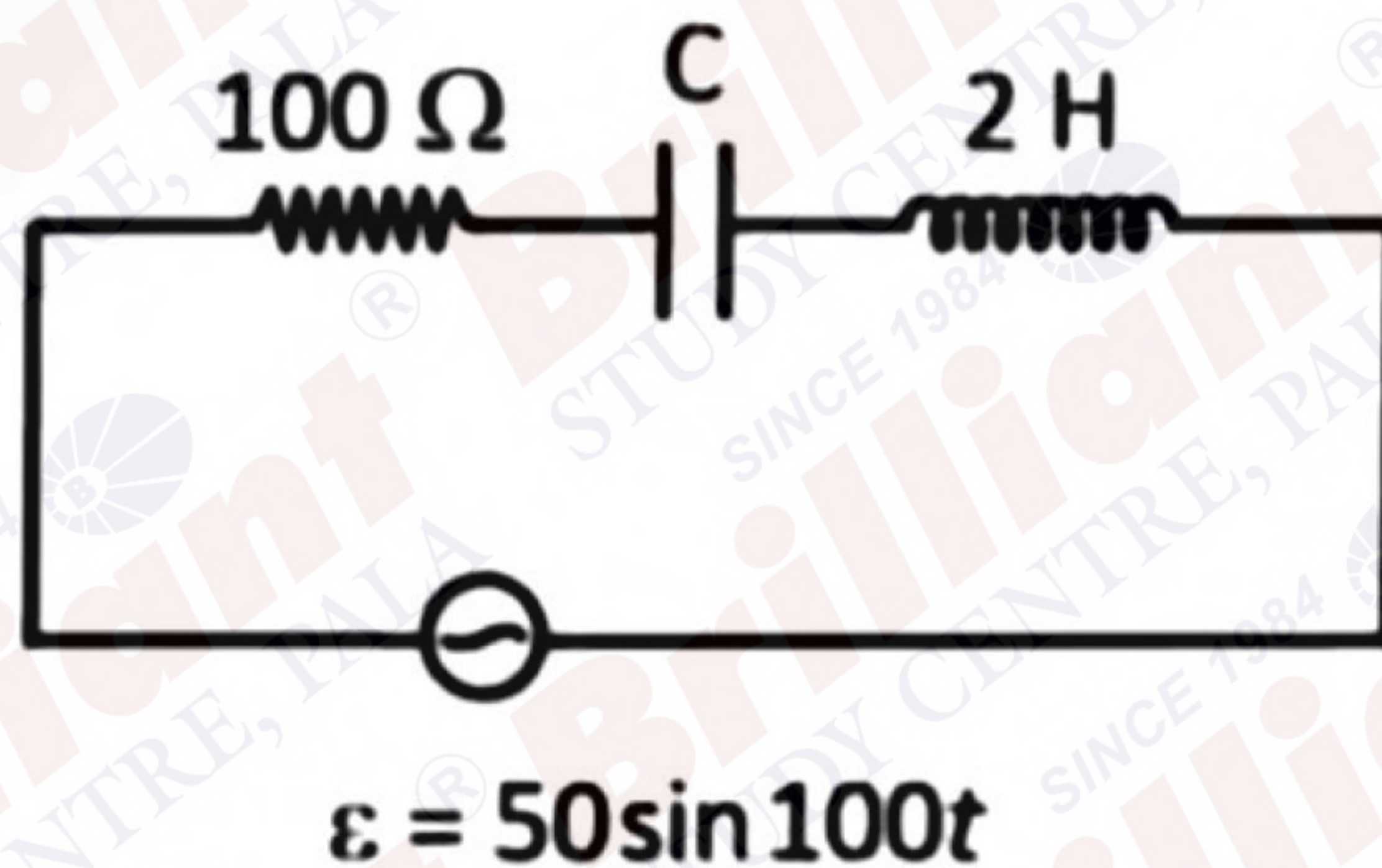


- 1) 1/4 2) 1/2 3) 1/8 4) 1

QN Two point particles of masses $2m$ and m are attached to ends of a massless rod of length l . Find Angular momentum of this system about an axis passing through their center of mass and perpendicular to the rod if the system is rotating with angular velocity ω about the axis.

- 1) $\frac{2}{3}m\omega l^2$ 2) $\frac{1}{3}m\omega l^2$ 3) $\frac{m\omega l^2}{9}$ 4) $m\omega l^2$

QN The peak current in given LCR series AC circuit shown is 5 mA, then capacitance in μF is nearly.

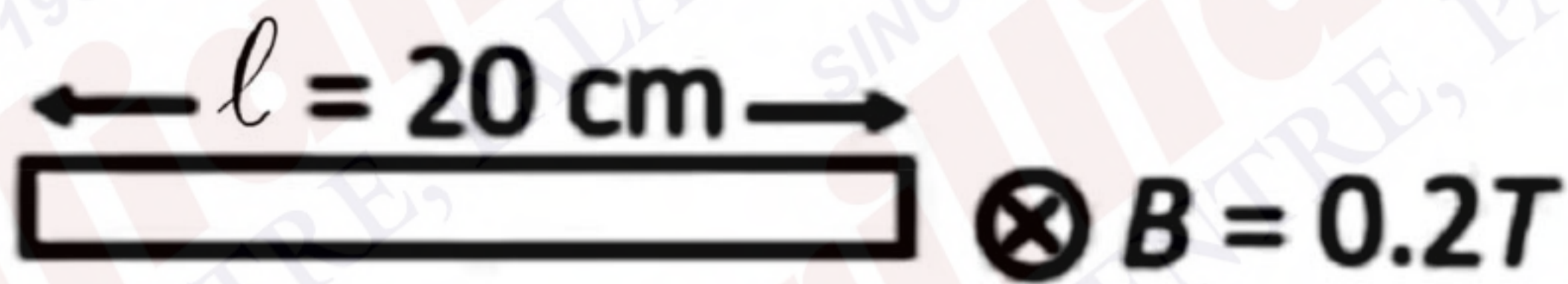


- 1) 1 2) 3 3) 4 4) 5

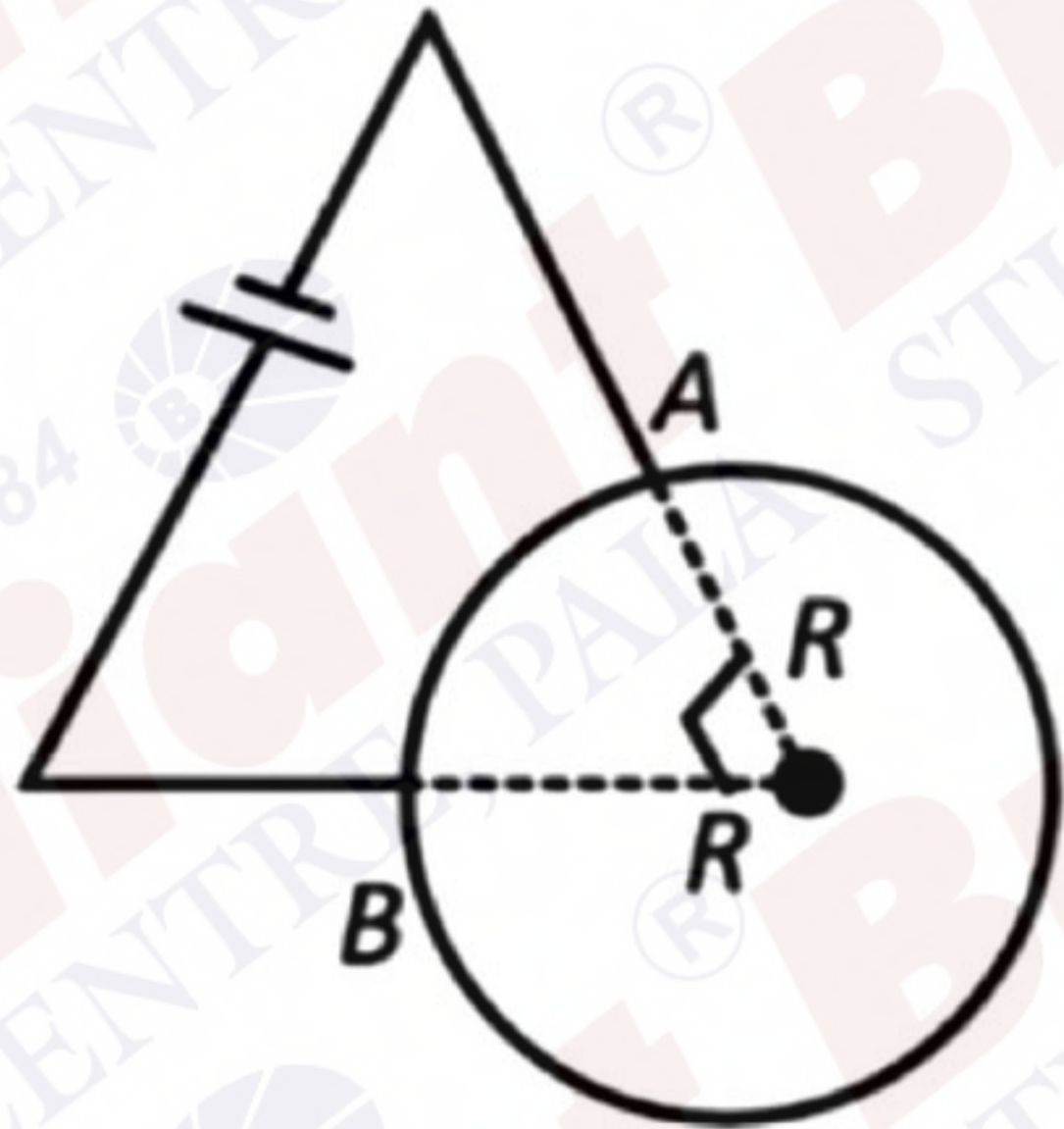
QN Two media of refractive indices n_1 & n_2 have plane interface. Speed of light in 1st is $2.4 \times 10^8 \text{ m/s}$ & in 2nd is $2.8 \times 10^8 \text{ m/s}$. Find critical angle of incidence when light travels from 1st to 2nd.

- 1) $\sin^{-1}\left(\frac{6}{7}\right)$ 2) $\sin^{-1}\left(\frac{7}{8}\right)$ 3) $\sin^{-1}\left(\frac{8}{9}\right)$ 4) $\tan^{-1}\left(\frac{9}{8}\right)$

QN Find emf induced in rod after it has fallen through 2000 m under gravity in volt.



QN A uniform wire is bent into the shape of a circle of radius R has resistance per unit length $= \lambda$. A battery is connected across point A and B which subtends an angle of 90° at its centre as shown. Find the equivalent resistance of this circuit.



- 1) $\frac{3\pi}{4}\lambda R$ 2) $\frac{\pi}{2}\lambda R$ 3) $\frac{3\pi}{8}\lambda R$ 4) $\frac{\pi}{4}\lambda R$

QN A square frame is formed by joining four identical solid cylinders _____ as its sides. Each cylinder has length L , mass M , and radius R . Find the moment of inertia of the frame about the axis lying in the plane of the frame and joining the centers of two opposite cylinders.

- 1) $I = \frac{2}{3}ML^2 + \frac{3}{2}MR^2$ 2) $I = \frac{1}{3}ML^2 + \frac{3}{2}MR^2$
 3) $I = \frac{2}{3}ML^2 + \frac{1}{2}MR^2$ 4) $I = \frac{1}{3}ML^2 + \frac{1}{2}MR^2$

QN A bob of mass m is released as shown when it was making 60° with the vertical. Another identical mass 'm' kept at rest. Elastic collision occurs. Find R for completing circle

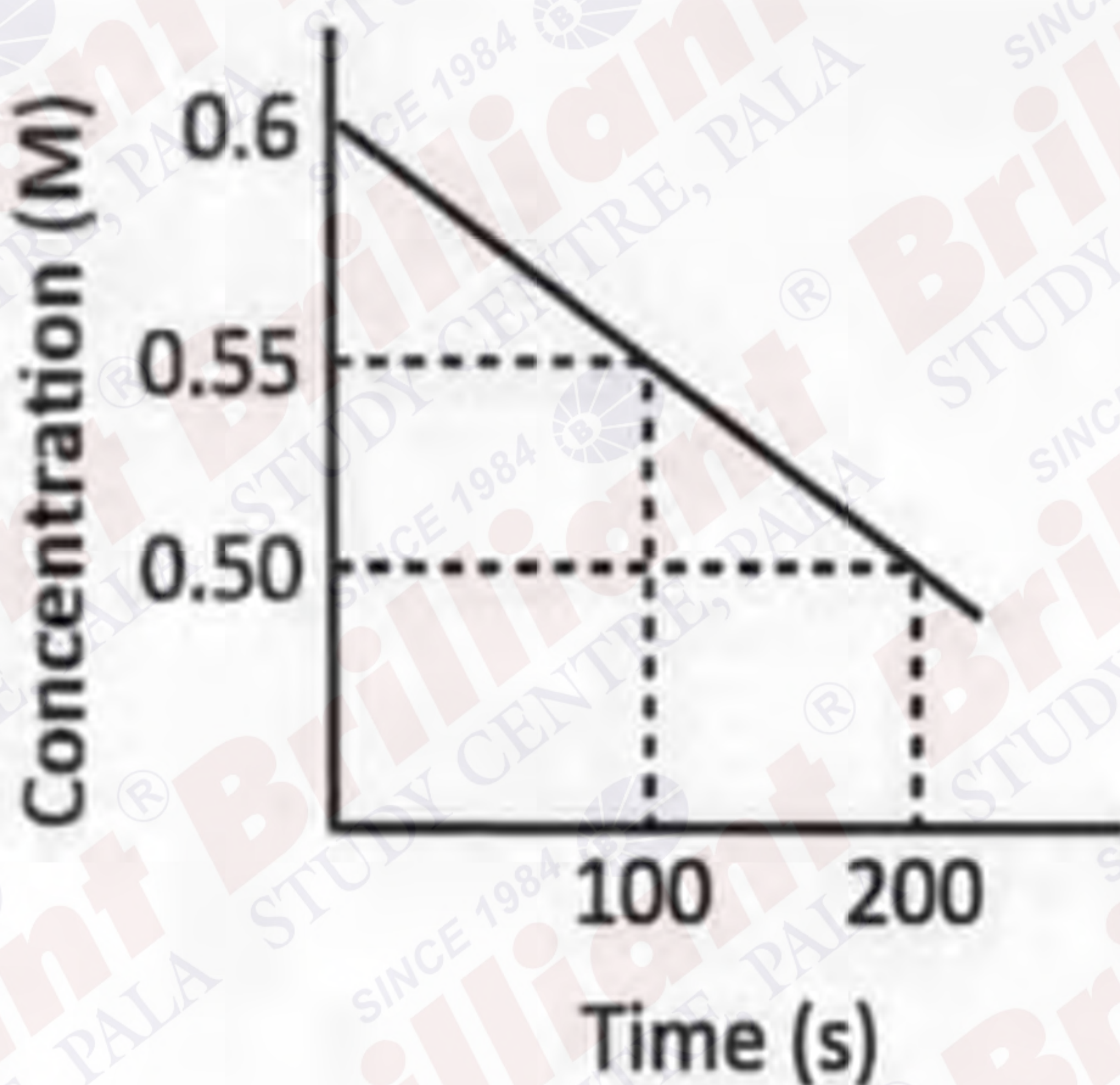


QN Statement-I : Sublimation is a purification technique that is used to separate those solid substances which changes from solid to vapour state without passing through liquid state.

Statment-II : If external atmospheric pressure is reduced, then boiling point of substance is decreased

- 1) Both statement-I and statement-II are correct
- 2) Both statement-I and statement-II are incorrect
- 3) Statement-I is correct and statement-II is incorrect
- 4) Statement-I is incorrect and statement-II is correct

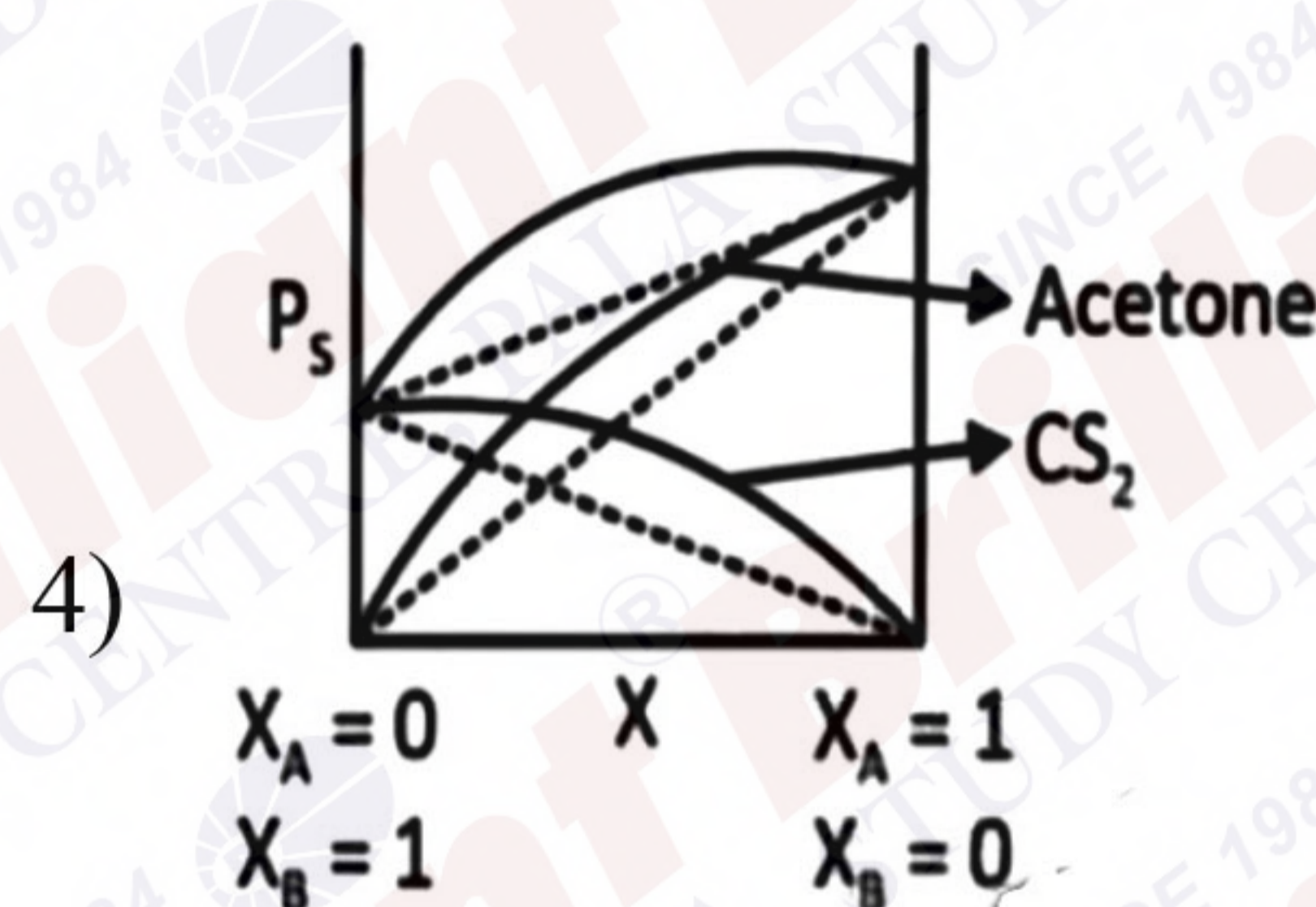
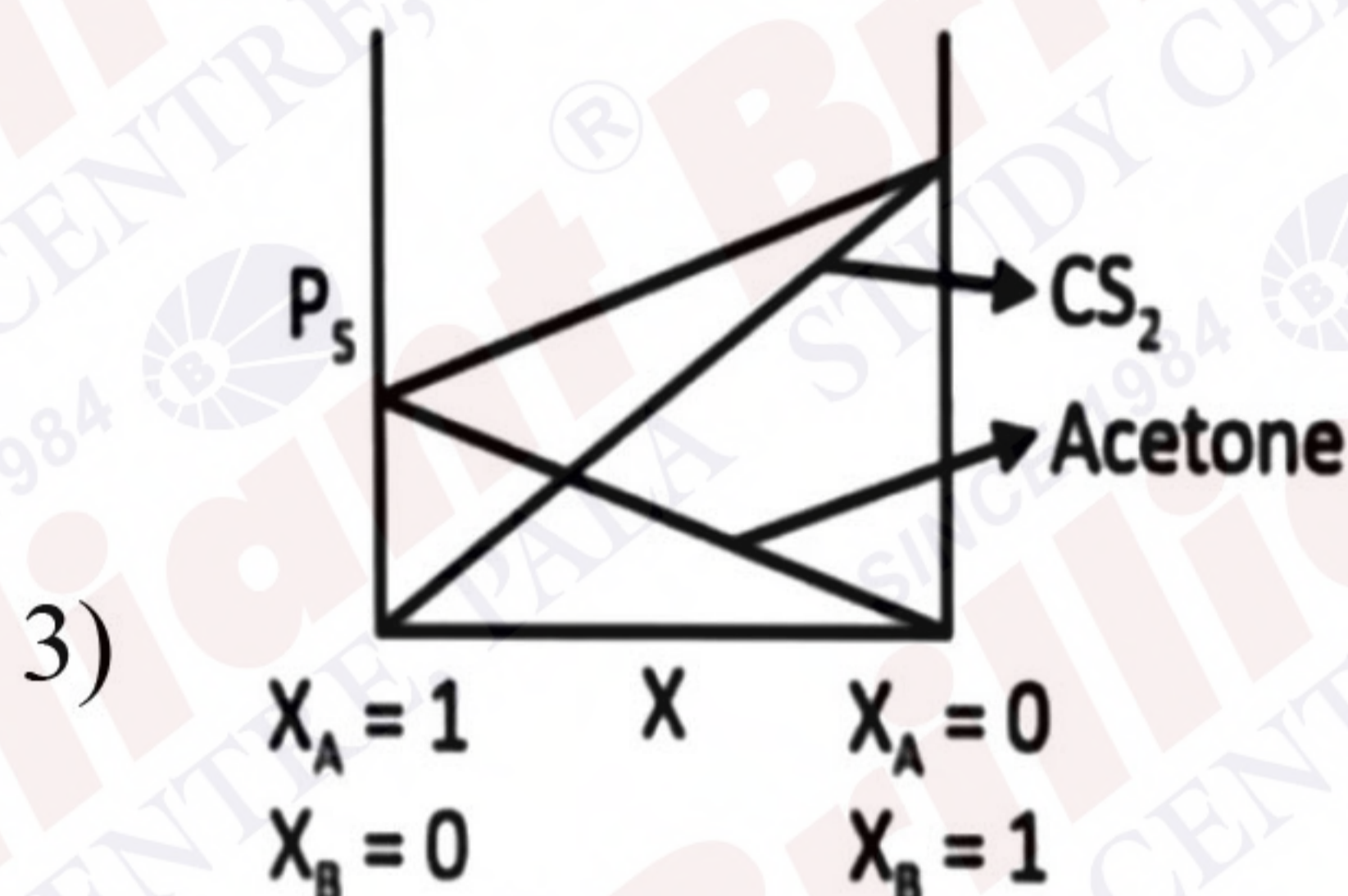
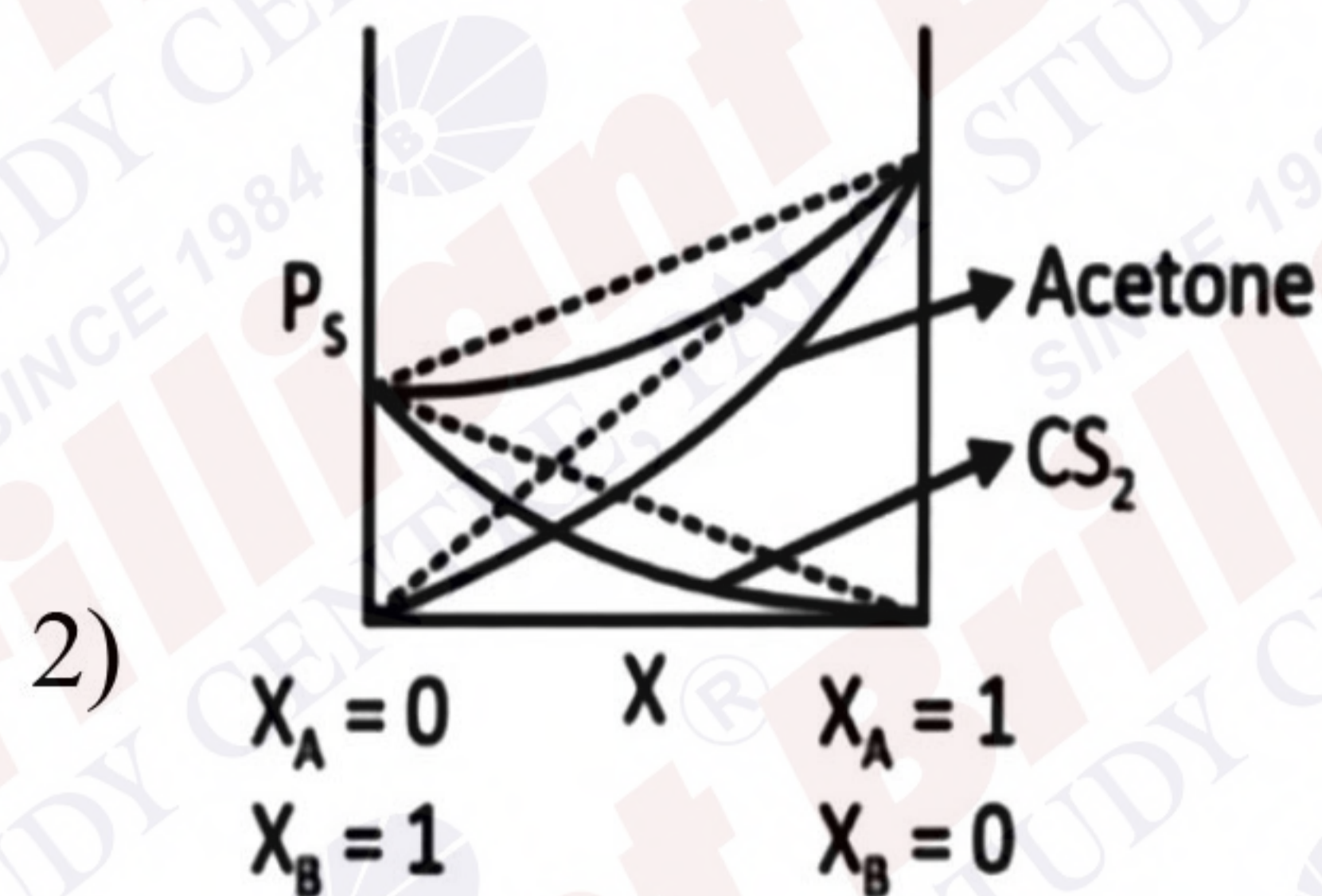
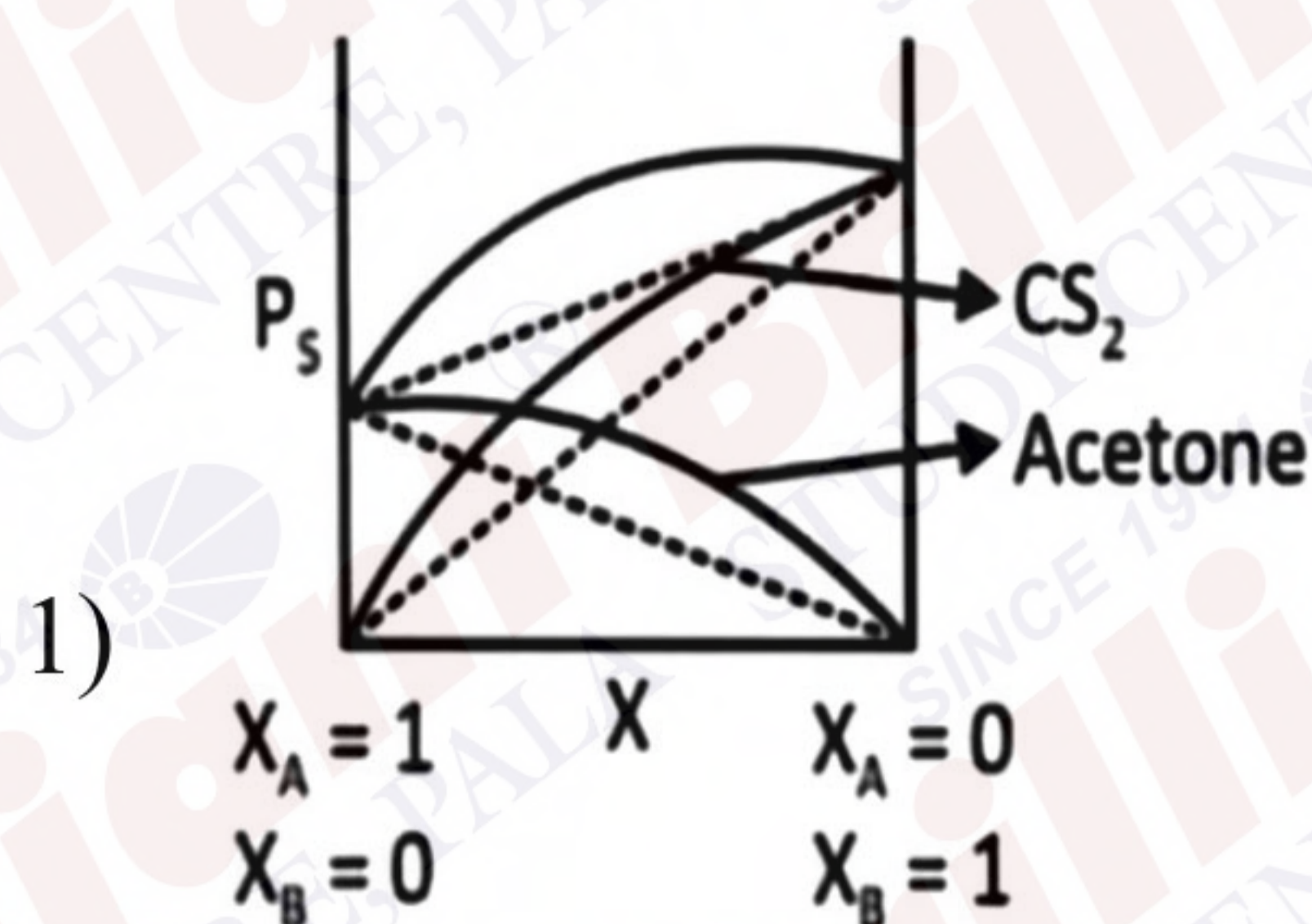
QN Consider the following graph of concentration vs time



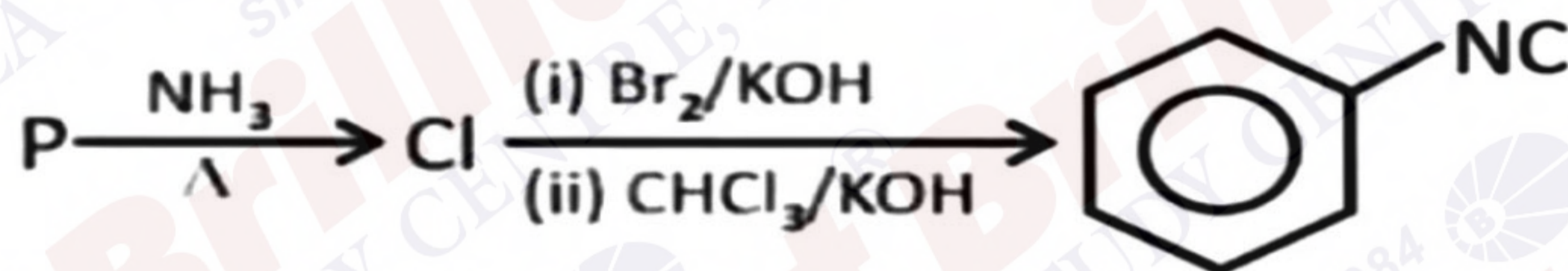
Find half-life fo reaction

- 1) 600 s
- 2) 200 s
- 3) 300 s
- 4) 100 s

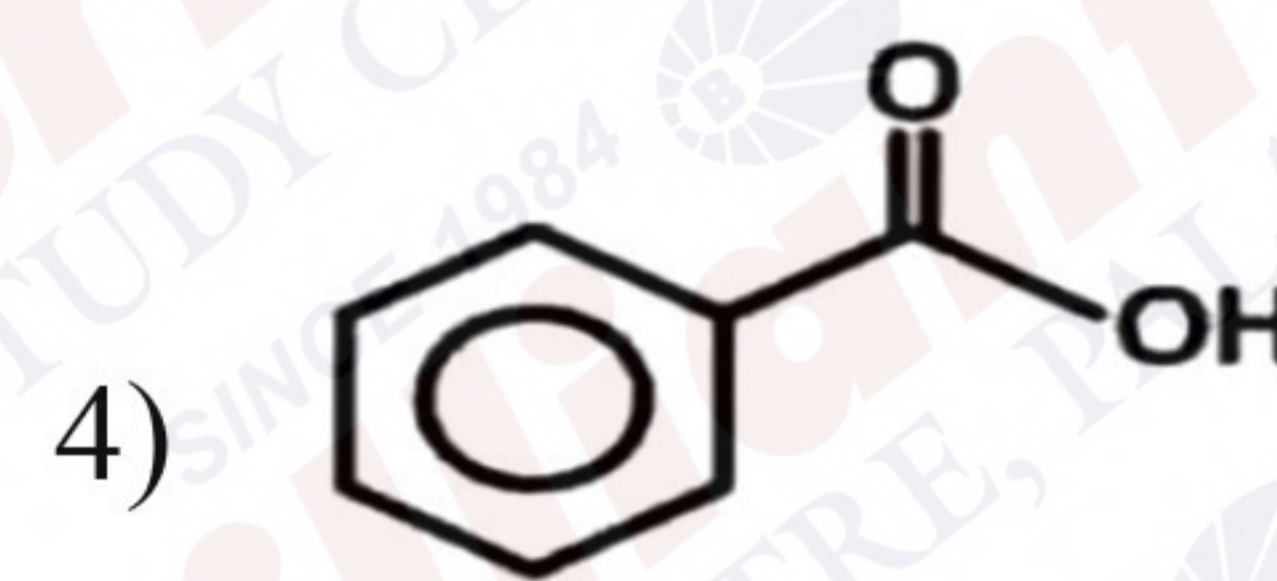
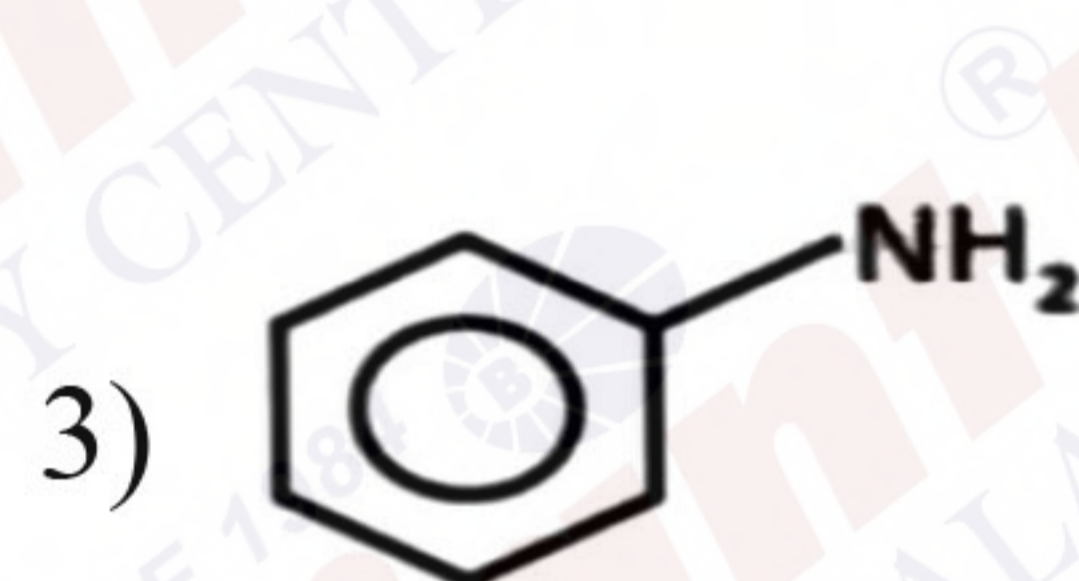
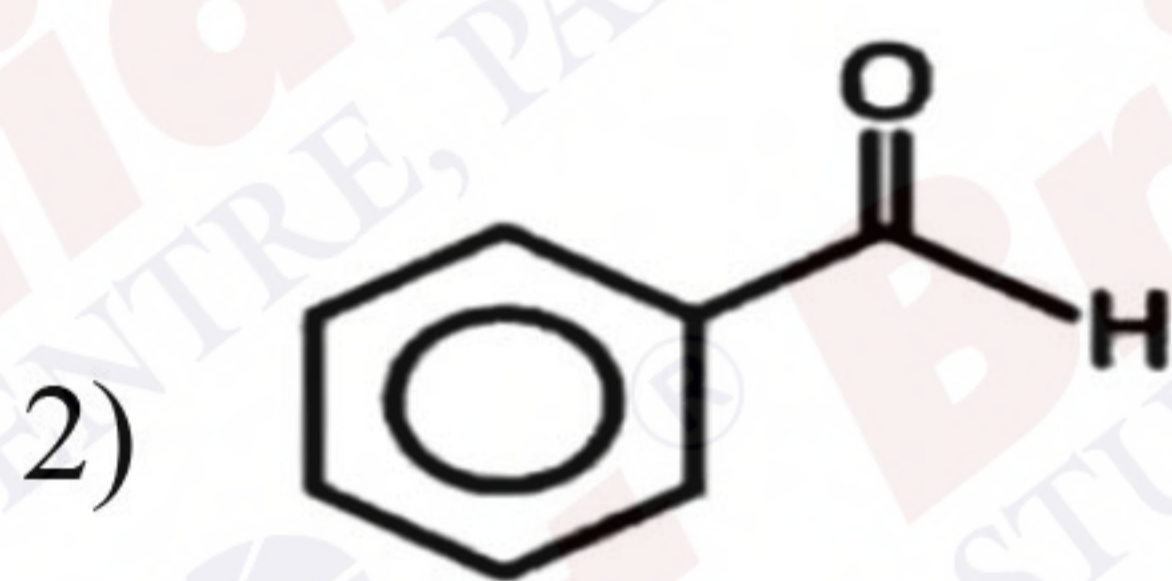
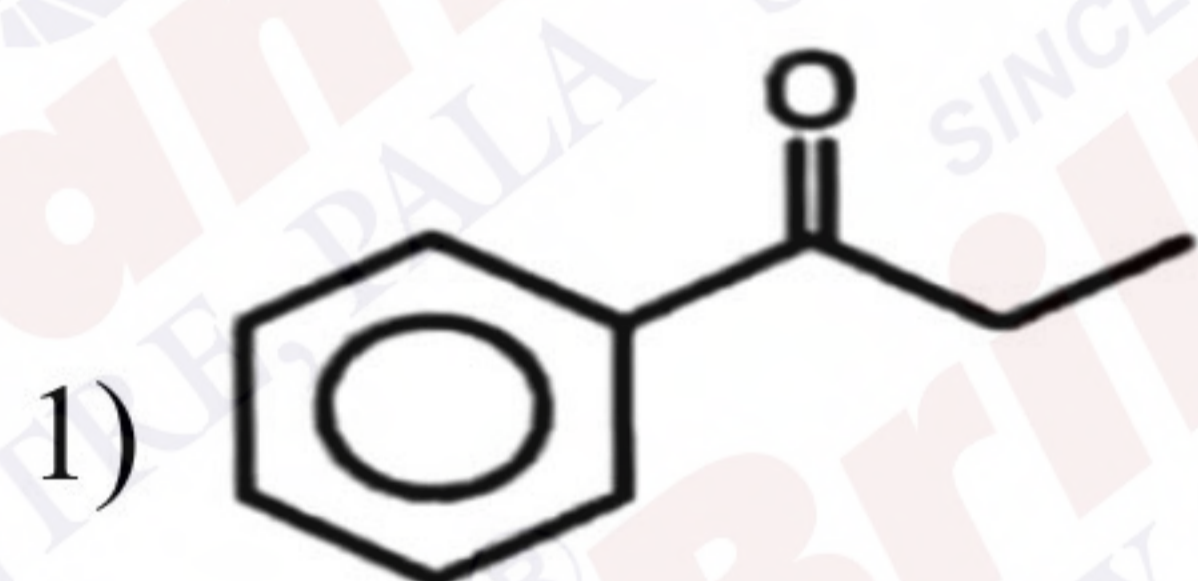
QN A binary solution is formed by mixing acetone (A) and CS₂(B). The variation of vapour pressure vs mole fraction will be



QN Consider the following reaction :



The structure of 'P' is



QN Consider the two complexes



Find the ratio of CFSE of I to II complex (neglect pairing energy and consider Δ₀ for both complexes to be x)

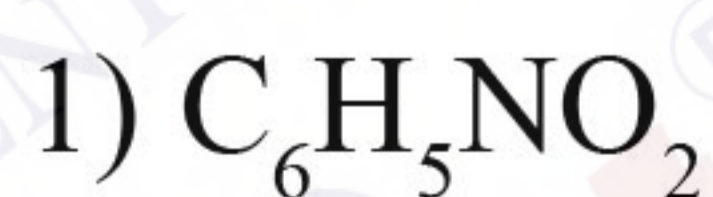
1) 2

2) 1/2

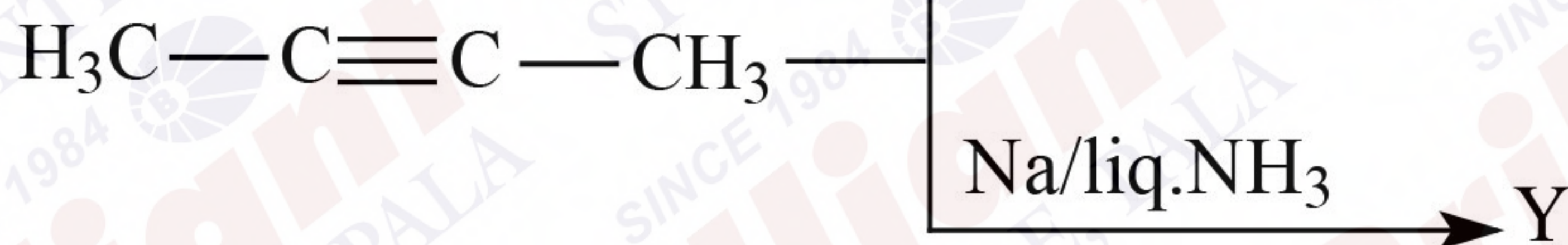
3) 1/3

4) 2/3

QN Which of the following undergo nitration at fastest rate?



QN



- 1) Both X and Y are stereoisomer
- 2) Boiling point of X is lesser than Y
- 3) Dipole moment of X is zero
- 4) They give same product on ozonolysis

QN

For the following change, $\text{H}_2\text{O}_{(\ell)} \xrightarrow{5^\circ\text{C}} \text{H}_2\text{O}_{(\text{g})} \xrightarrow{100^\circ\text{C}}$. Select the correct answer

- 1) $q = +ve$; $w = +ve$; $\Delta H = +ve$
- 2) $q = -ve$; $w = -ve$; $\Delta H = +ve$
- 3) $q = +ve$; $w = -ve$; $\Delta H = +ve$
- 4) $q = -ve$; $w = -ve$; $\Delta H = -ve$

QN

The correct order of ionisation energy of Cl, S, P, Al, Si is

- 1) $\text{Cl} > \text{P} > \text{S} > \text{Si} > \text{Al}$
- 2) $\text{P} > \text{Cl} > \text{S} > \text{Al} > \text{Si}$
- 3) $\text{Cl} > \text{S} > \text{P} > \text{Si} > \text{Al}$
- 4) $\text{Cl} > \text{Al} > \text{Si} > \text{P} > \text{S}$

QN

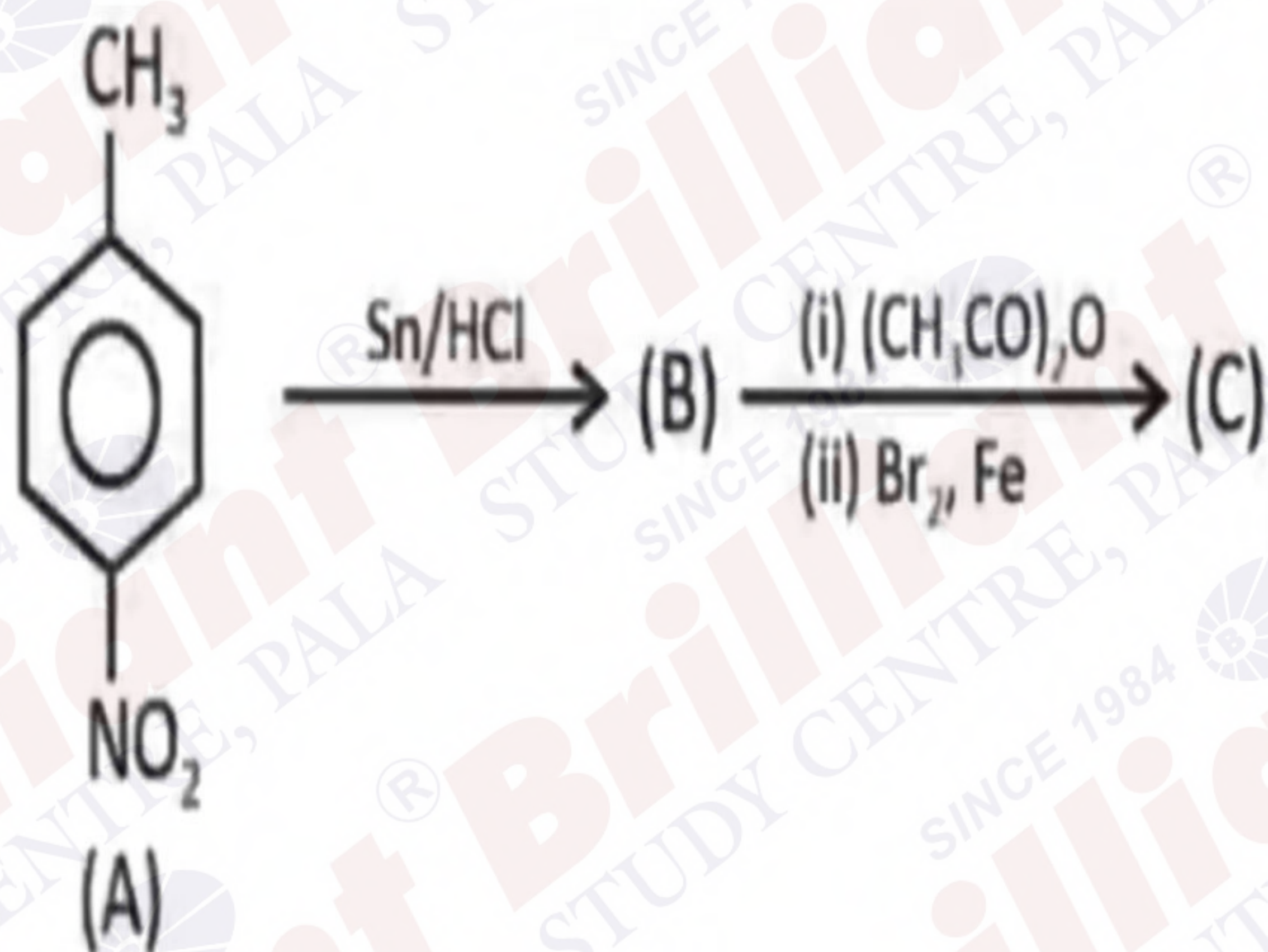
Statement-I : $[\text{CoBr}_4]^{2-}$ absorbs lesser energy than $[\text{CoCl}_4]^{2-}$

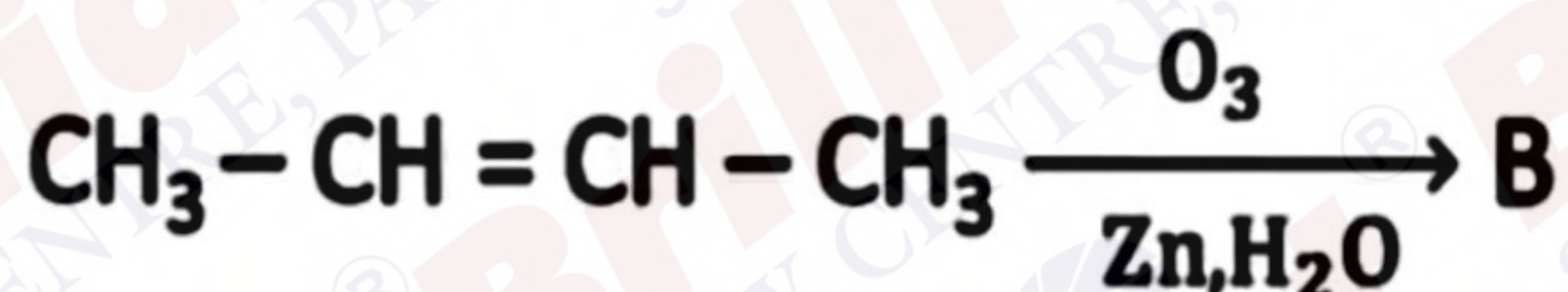
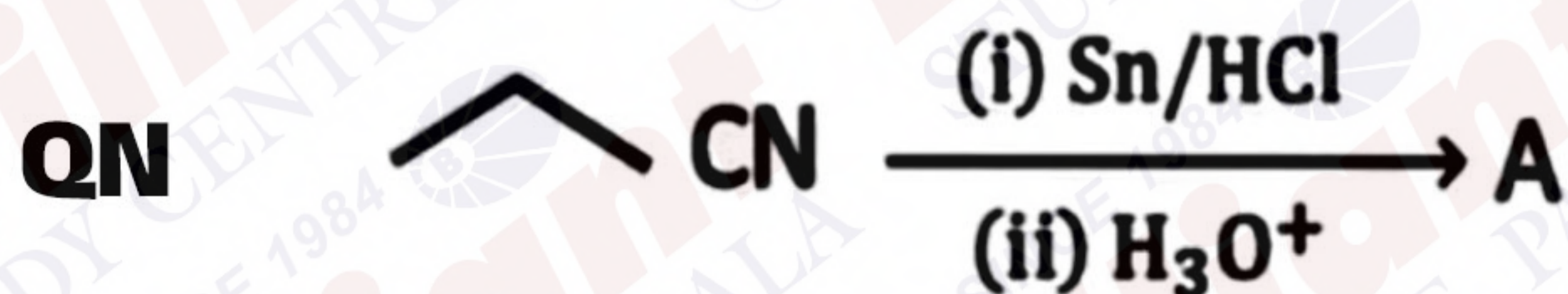
Statement-II : $[\text{CoCl}_4]^{2-}$ has higher crystal field splitting energy than $[\text{CoBr}_4]^{2-}$

- 1) Both statement-I and statement-II are correct
- 2) Both statement-I and statement-II are incorrect
- 3) Statement-I is correct and statement-II is incorrect
- 4) Statement-I is incorrect and statement-II is correct

QN

In the reaction sequence, what is the mass (in grams) of product (C) formed?





A and B are mixed and treated with dil. base to give mixture of products.
Choose the incorrect product.



QN Correct statement about 13th group

- A) Electronegativity decreases regularly down the graph
B) Ionic radii decreases down the graph
C) Boron has highest ionisation energy
D) Trichloride of aluminium are covalent in nature

- 1) A and C only 2) C, D only 3) A, C, D only 4) B, C, D only

QN Match list-I with list-II

	List-I		List-II
A.	Bayer's unsaturation test	(I)	Violet/purple colour
B.	Cerric ammonium nitrate test of alcohols	(II)	Red colour
C.	Tollen's reagent test	(III)	Silver mirror obtained
D.	FeCl ₃ test of phenol	(IV)	Pink colour discharge

- 1) A-II; B-I; C-IV; D-III 2) A-II; B-I; C-III; D-IV
3) A-IV; B-II; C-III; D-I 4) A-IV; B-III; C-II; D-I

QN Consider the given cell $\text{Ag(s)}|\text{AgCl(s)}|\text{Cl}^-(\text{aq})|\text{FeCl}_2(\text{aq}), \text{FeCl}_3(\text{aq})|\text{Pt(s)}$. In which of the following cases, E_{cell} will increase

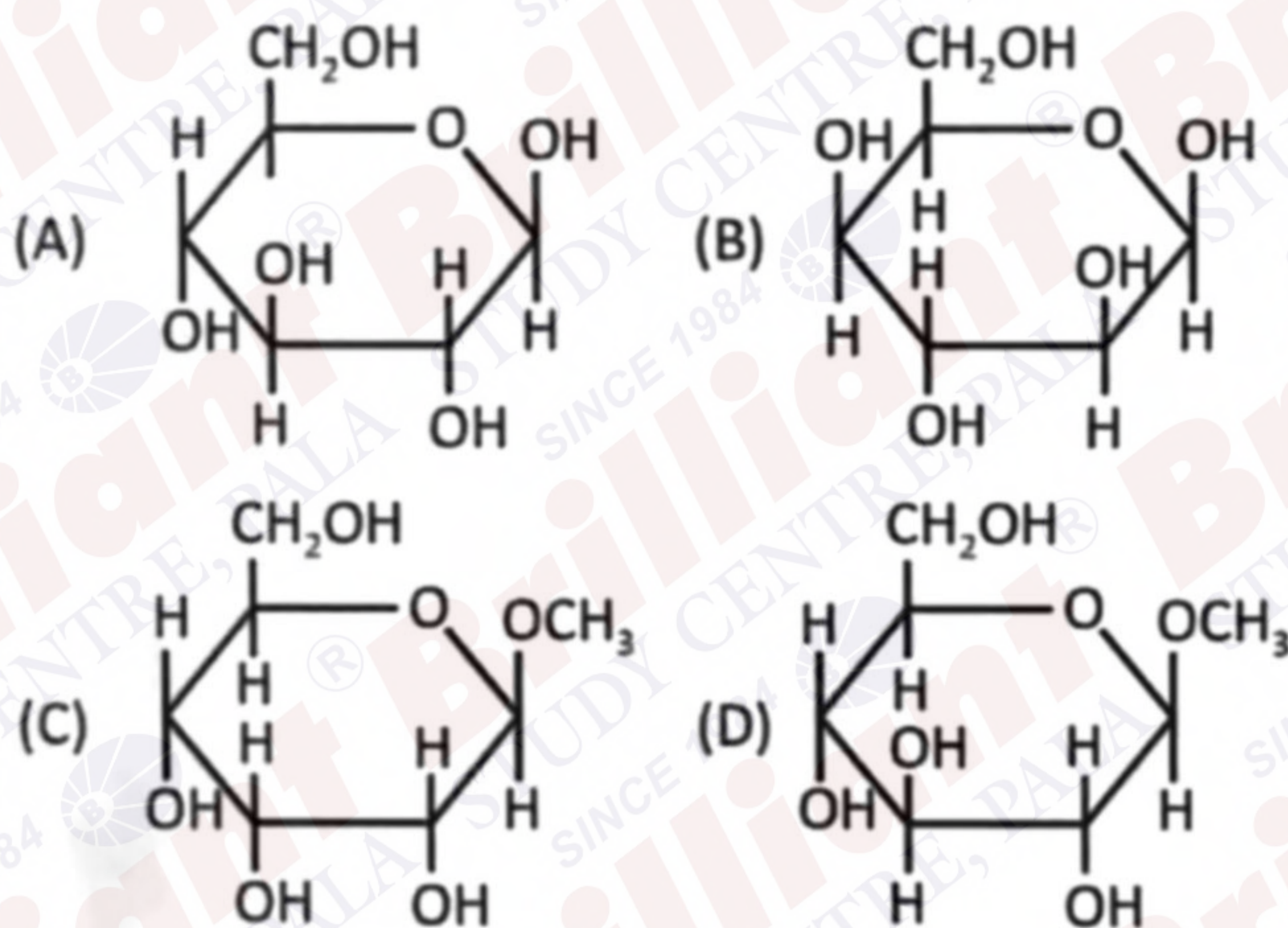
- 1) [Fe²⁺] increases, [Cl⁻] increases 2) [Fe²⁺] increases, [Cl⁻] decreases
3) [Fe³⁺] and [Cl⁻] increases 4) [Fe²⁺] decreases, [Fe³⁺] decreases

QN Consider the following statements and choose the correct option(s)

- A) Ni²⁺ forms pink colour complex with dimethyl glyoximate ligand (dmg)
B) Ni²⁺ complex with dmg contains two ring of five membered
C) Ni²⁺ has 2 unpaired electron in d-orbitals in the complex
D) Ni²⁺ complex with dmg is soluble at pH = 9

- 1) Only A & B 2) Only B & C 3) Only A & C 4) Only A, C and D

QN Consider the following molecules



The examples of non-reducing sugar(s) are

- 1) A, B only 2) A, C only 3) B, D only 4) C, D only

QN Following molecules are given: HNO_3 , NF_3 , H_2SO_4 , O_3 . Consider the molecule (T) having maximum number of lone pairs (on all atoms). The bond angle ($\angle\text{XMX}$), where M is central atom in T is

- 1) 110° 2) 97° 3) 102° 4) 115°

QN Match the list-I with list-II and choose the correct option

	List-I		List-II
(a)	2 nd orbit of He^+ ion	(i)	$-1.96 \times 10^{-17} \text{ J}$
(b)	3 rd orbit of H-atom	(ii)	$-2.42 \times 10^{-19} \text{ J}$
(c)	1 st orbit of Li^{2+} ion	(iii)	$-2.178 \times 10^{-18} \text{ J}$
(d)	2 nd orbit of Li^{2+} ion	(iv)	$-4.9 \times 10^{-18} \text{ J}$

- 1) a-III; b-II; c-I; d-IV 2) a-III; b-II; c-IV; d-I
 3) a-IV; b-III; c-II; d-I 4) a-I; b-II; c-III; d-IV

QN $\text{C}_5\text{H}_{11}\text{Br}$ reacts with aq. KOH without rearrangement. How many optically active compounds are formed

QN xg of pure Cl_2 is reacted with $\text{Ba}(\text{OH})_2$ to form $\text{Ba}(\text{ClO}_3)_2 \cdot \text{Ba}(\text{OH})_2$ concentration is 1M and volume is 25 mL. Find x

QN The value of $\frac{{}^{100}C_{50}}{51} + \frac{{}^{100}C_{51}}{52} + \dots + \frac{{}^{100}C_{100}}{101}$ is

- 1) $\frac{2^{100}}{100}$ 2) $\frac{2^{101}}{101}$ 3) $\frac{2^{100}}{101}$ 4) $\frac{2^{101}}{100}$

QN The number of solutions of $13 \cos 2\theta + 8 \cos \theta - 3\sqrt{3} = 0$ if $\theta \in [-3\pi, 2\pi]$ is

- 1) 4 2) 3 3) 0 4) 5

QN If coefficient of x, x^2 and x^3 are in arithmetic progression of the binomial expansion of $(1+x^2)^2(1+x)^n$, $n \in \mathbb{N}$. Then sum of all values of 'n' is

QN Let $f(x) = \int \frac{e^x(2-x^2)}{\sqrt{1+x(1-x)^{\frac{3}{2}}}} dx$, with $f(0)=0$, then $f\left(\frac{1}{2}\right)$ is

- 1) $\sqrt{2e} + 1$ 2) $\sqrt{2e} - 1$ 3) $\sqrt{3e} + 1$ 4) $\sqrt{3e} - 1$

QN $\int_{\frac{\pi}{24}}^{\frac{5\pi}{24}} \frac{dx}{1 + \sqrt[3]{\tan x}}$ =

QN Find the area bounded by $y = \max\{\sin x, \cos x\}$ where $x \in \left[0, \frac{3\pi}{2}\right]$ with x-axis

- 1) 3 2) 3π 3) 4π 4) 4

QN If $S = \{1, 2, 3, \dots, 100\}$. Select two different numbers m and n such that $n - m \geq 10$ having probability $\frac{a}{b}$. Find $a+b$ where a and b are co-prime

QN A rectangle informed by lines $x = 0, y = 0, x = 3$ and $y = 4$. A line \perp^r to $3x+4y+6=0$ divides the rectangle into two equal parts then distance of the line from $\left(-1, \frac{3}{2}\right)$ is

QN For the differential equations $x^4 dy + (4x^3 y + \sin x) dx = 0$, it is given that $y\left(\frac{\pi}{2}\right) = 0$. Then the value of $x^4 y\left(\frac{\pi}{3}\right)$ is equal to

- 1) $\frac{9}{2}$ 2) $\frac{81}{2}$ 3) $\frac{27}{2}$ 4) $\frac{243}{2}$

QN For given vectors $\vec{a} = -\hat{i} + \hat{j} + 2\hat{k}$ and $\vec{b} = 2\hat{i} - \hat{j} + \hat{k}$ where $\vec{c} = \vec{a} \times \vec{b}$ and $\vec{d} = \vec{c} \times \vec{b}$. Then the value of $(\vec{a} - \vec{b}) \cdot \vec{d}$ is

- 1) -35 2) 53 3) -52 4) 25

QN Number of 4 letters words with or without meaning formed from the letters of the word PQRSSSTTUVV is

- 1) 1232 2) 1400 3) 1422 4) 1162

QN Let $A = \{-2, -1, 0, 1, 2, 3, 4\}$ and R be a relation such that

$$R = \{(x, y) : 2x + y \leq -2, x \in A, y \in A\}$$

Let ℓ = number of elements in R, m = minimum number of elements to be added in R to make it reflexive, n = minimum number of elements to be added in R to make it symmetric, then $(\ell + m + n)$ is

QN If A is matrix of order 3 and $x = |3\text{adj}(A^2) \cdot \text{adj}(2A)|$ and $|A| = 6$ and $x = 2^n \times 3^m$ then $m+n$ is

- 1) 21 2) 25 3) 27 4) 19

QN If point B and C lies on the line $\frac{x}{1} = \frac{1-y}{-2} = \frac{z-2}{3}$ and point A is (1,6,3). If $BC = 10$, then the area of ΔABC is

- 1) $2\sqrt{13}$ 2) $5\sqrt{13}$ 3) $6\sqrt{13}$ 4) $4\sqrt{13}$

QN Let the domain of $f(x) = \log_3 \log_5 \log_7 (9x - x^2 - 13)$ is (m, n) . Let $\frac{n}{3}$ and $\frac{3m}{8}$ be eccentricity and

length of latus rectum of hyperbola $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ respectively, then the value of $\left(\frac{a+b^2}{a-b^2}\right)$ is equal

- to
- 1) 7 2) 9 3) 3 4) 13

